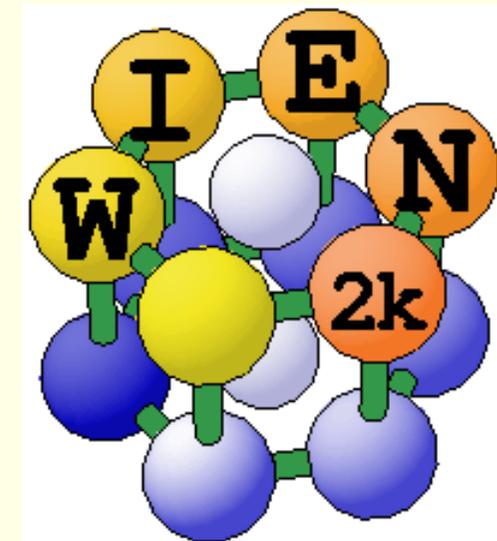


Calculations of NMR chemical shifts in solids

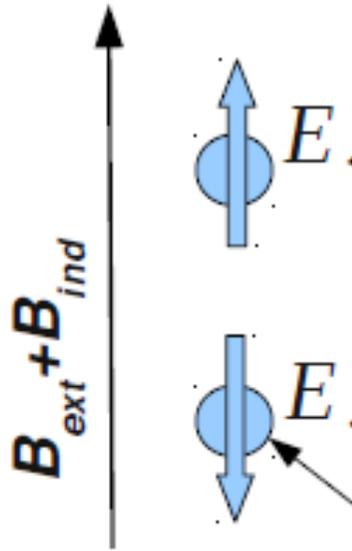
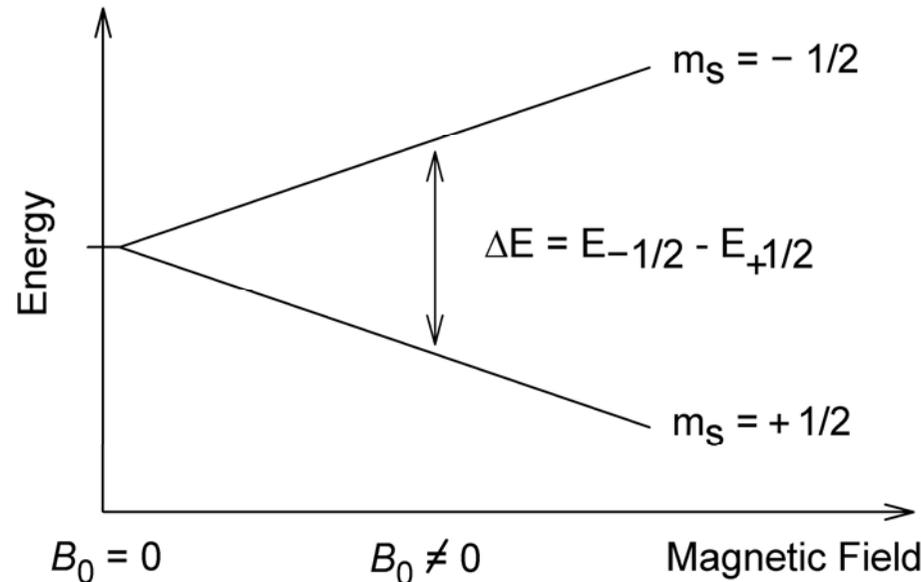
Peter Blaha

Institute of Materials Chemistry
TU Vienna, Austria

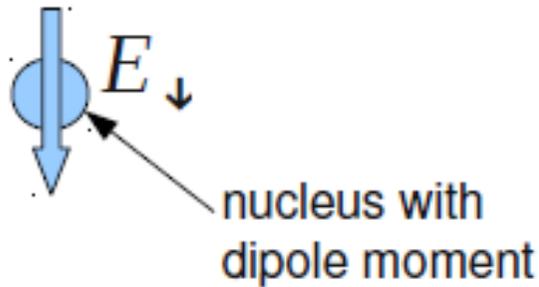




NMR spectroscopy



$$\Delta E = -\gamma m \hbar (B_{ext} + B_{ind})$$



$B_{ext} + B_{ind}$ is measured at any nucleus by detecting transition energy related to reorientation of its magnetic moment



NMR Hamiltonian



perturbation

$$H_{NMR} = H_Z + H_\sigma + H_Q + H_D + H_J + \dots$$

$$H_Z = -\mu \cdot B_{ext}$$

Zeeman Hamiltonian

electric quadrupole
interaction (EFG)

Indirect spin-
spin coupling

direct dipolar
coupling

$$H_\sigma = -\mu \cdot B_{ind}$$

magnetic shielding



NMR Hamiltonian



$$H_{NMR} = H_Z + H_\sigma + H_Q + H_D + H_J + \dots$$

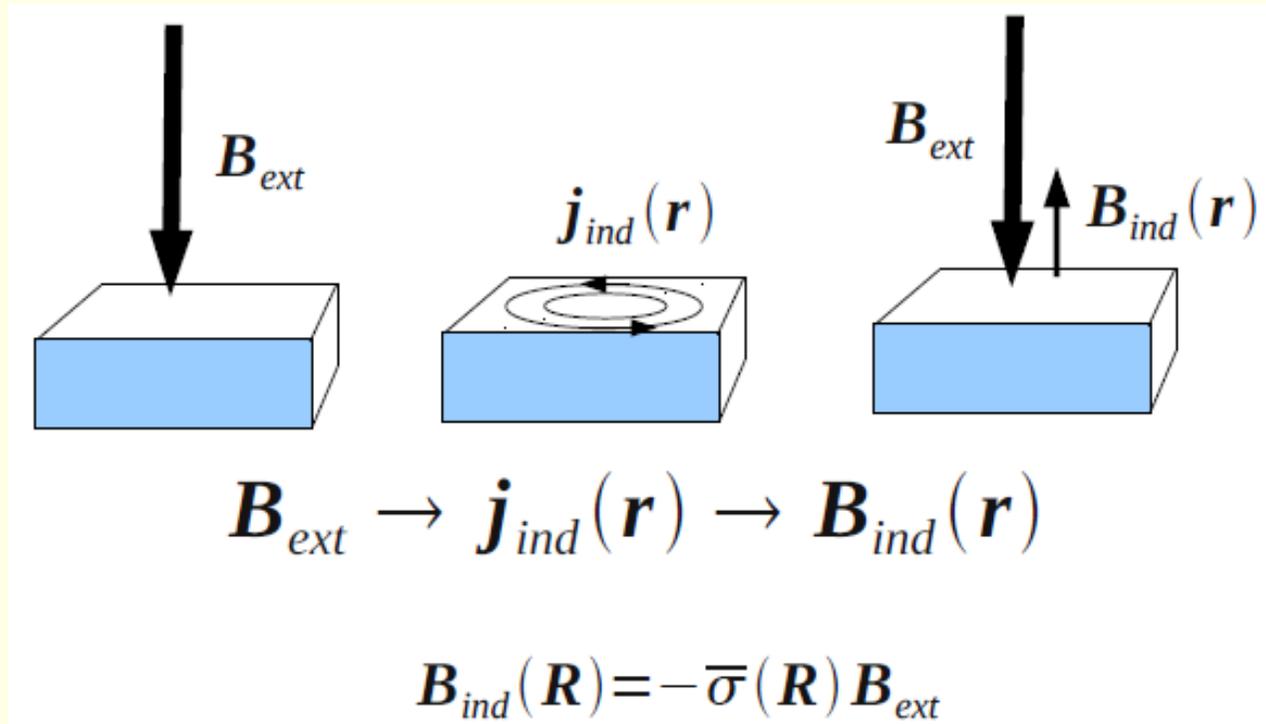
$H_Z = -\mu \cdot B_{ext}$
Zeeman Hamiltonian

$H_\sigma = -\mu \cdot B_{ind}$
magnetic shielding

$H_Q \approx eQ\Phi / h$
quadrupole interaction



NMR shielding, chemical shift:



$\sigma(\mathbf{R})$ is the **shielding tensor** at the nucleus \mathbf{R}

chemical shift:

$$\delta(\text{ppm}) = \frac{\sigma_{ref} - \sigma}{1 - \sigma_{ref}} \times 10^6$$



Biot - Savart law



The induced **magnetic field** (\mathbf{B}_{ind}) is derived from the **induced current** (\mathbf{j}_{ind}) using a standard formula:

$$\mathbf{B}_{ind}(\mathbf{r}) = \frac{1}{c} \int d^3 r' \mathbf{j}(\mathbf{r}') \times \frac{\mathbf{r} - \mathbf{r}'}{|\mathbf{r} - \mathbf{r}'|^3}$$

in DFT the **current density** $\mathbf{j}(r)$ will be:

$$\mathbf{j}_{ind}(\mathbf{r}') = \sum_o \left[\langle \Psi_o^{(1)} | \mathbf{J}^{(0)}(\mathbf{r}') | \Psi_o^{(0)} \rangle + \langle \Psi_o^{(0)} | \mathbf{J}^{(0)}(\mathbf{r}') | \Psi_o^{(1)} \rangle \text{ diamagn.} \right. \\ \left. + \langle \Psi_o^{(0)} | \mathbf{J}^{(1)}(\mathbf{r}') | \Psi_o^{(0)} \rangle \right], \quad (4) \text{ paramagn.}$$

perturbed w.f. Ψ^1 is obtained from perturbation theory

$$|\Psi_o^{(1)}\rangle = \sum_e |\Psi_e^{(0)}\rangle \frac{\langle \Psi_e^{(0)} | H^{(1)} | \Psi_o^{(0)} \rangle}{\epsilon_o - \epsilon_e}$$

sum over **all** empty states

$$H^{(1)} = \frac{1}{2c} \mathbf{r} \times \mathbf{p} \cdot \mathbf{B}$$

magnetic field



sum over **ALL** empty states:



$$|\Psi_o^{(1)}\rangle = \sum_e |\Psi_e^{(0)}\rangle \frac{\langle \Psi_e^{(0)} | H^{(1)} | \Psi_o^{(0)} \rangle}{\epsilon_o - \epsilon_e}$$

- standard APW basis set $u_l(r, E_l)$ only good near linearization E_l
- adding additional LOs at high energies (up to 1000 Ry !!!)
- $H^{(1)}$ contains the ∇ operator, so we need to represent the radial derivative of $u_l(r, E_l)$ at $l \pm 1$

- adding "NMR-los" $\xi_{l,k}(r, \tilde{\epsilon}) = \begin{cases} r \frac{d}{dr} u_{l+1}(r, \tilde{\epsilon}) + (l+2)u_{l+1}(r, \tilde{\epsilon}), & k=1 \\ r \frac{d}{dr} u_{l-1}(r, \tilde{\epsilon}) - (l-1)u_{l-1}(r, \tilde{\epsilon}), & k=2 \end{cases}$

- x_nmr -mode in1 [-focus nmr_atom] will set that up automatically



practical calculation



- run normal scf cycle
- `x_nmr_lapw -mode in1 [-focus 0]`
 - *view the resulting `*in1c_nmr` file*
- `x_nmr_lapw [-p]`

- *creates several directories (`nmr_q0`, `nmr_pqx`, `nmr_mqx`, `nmr_pqy`, ..) and performs `lapw1/2` steps for several k -meshes ($k \pm q$)*
- *creates the current*
- *integrates the current*

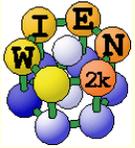
$$\mathbf{r} \cdot \hat{\mathbf{u}}_i = \lim_{q \rightarrow 0} \frac{1}{2q} \left(e^{iq\hat{\mathbf{u}}_i \cdot \mathbf{r}} - e^{-iq\hat{\mathbf{u}}_i \cdot \mathbf{r}} \right)$$

- `tail case.outputnmr_integ`

- `:NMRTOT003 ATOM: O 3 NMR (total/ppm) TRACE = -366.9034 ANISO = -264.8134 ASYM = -0.9504`

- for analysis one can calculate the shift from certain bands (energy range) only
 - `x_nm_lapw [-p] -noinit -emin xx [-emax yy]`

```
WFFIL EF=.533144859350 (WFFIL, WFPRI, ENFIL,
7.00 10 4 (R-MT*K-MAX; MAX L IN WF, V-NM
0.30 19 0 (GLOBAL E-PARAMETER WITH n OTH
0 -0.58576 0.002 CONT 1
0 4.80000 0.000 CONT 1
0 36.60000 0.000 CONT 1
0 66.66000 0.000 CONT 1
0 104.26000 0.000 CONT 1
0 149.26000 0.000 CONT 1
0 201.50000 0.000 CONT 1
```



Test of accuracy: Ar atom



- the current \mathbf{j} and chemical shielding σ of a **spherical atom** can be calculated “exactly” from the density $\rho(\mathbf{r})$ (no perturbation theory) by:

$$\mathbf{j}_\rho(\mathbf{r}') = -\frac{1}{2c}\rho(\mathbf{r}')\mathbf{B} \times \mathbf{r}'$$

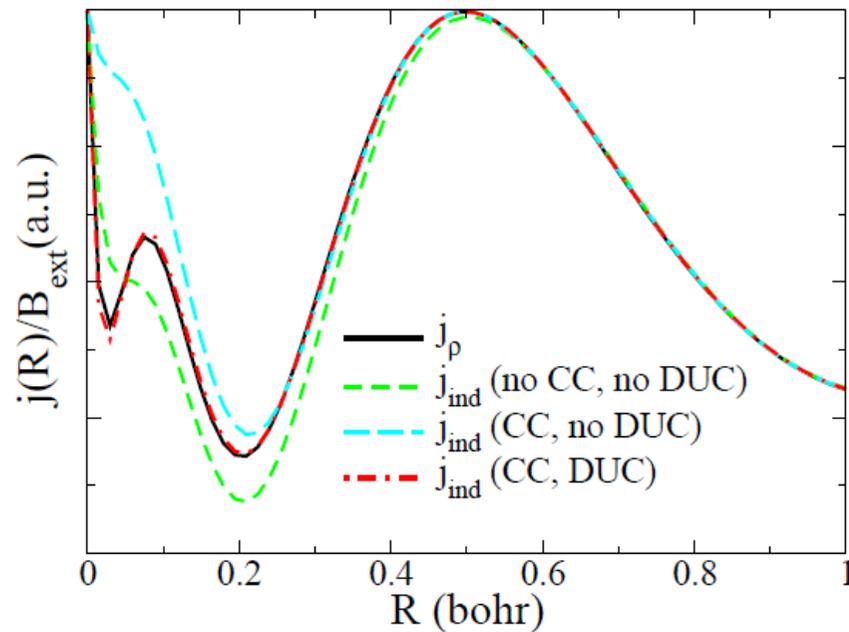
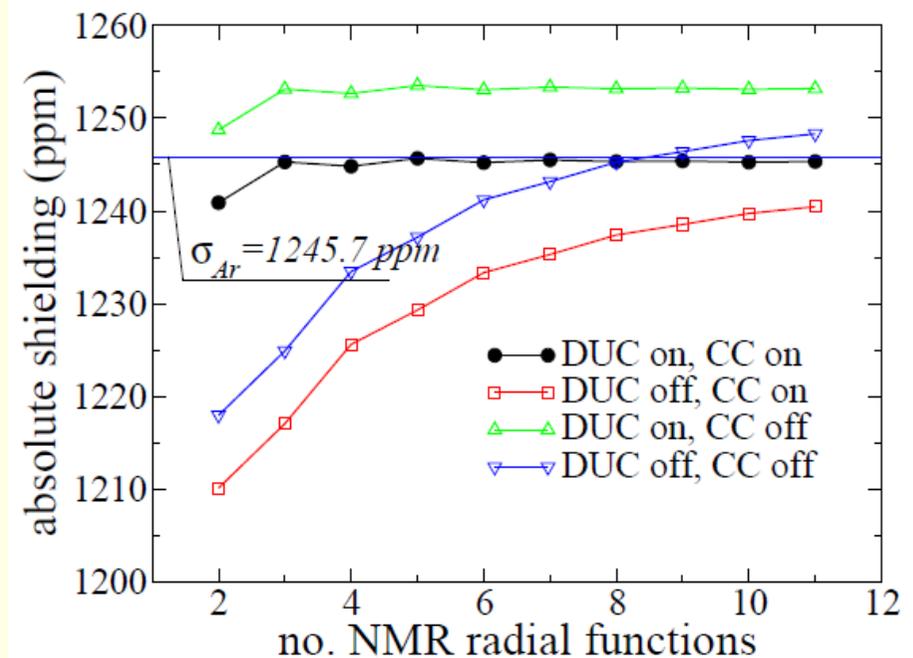


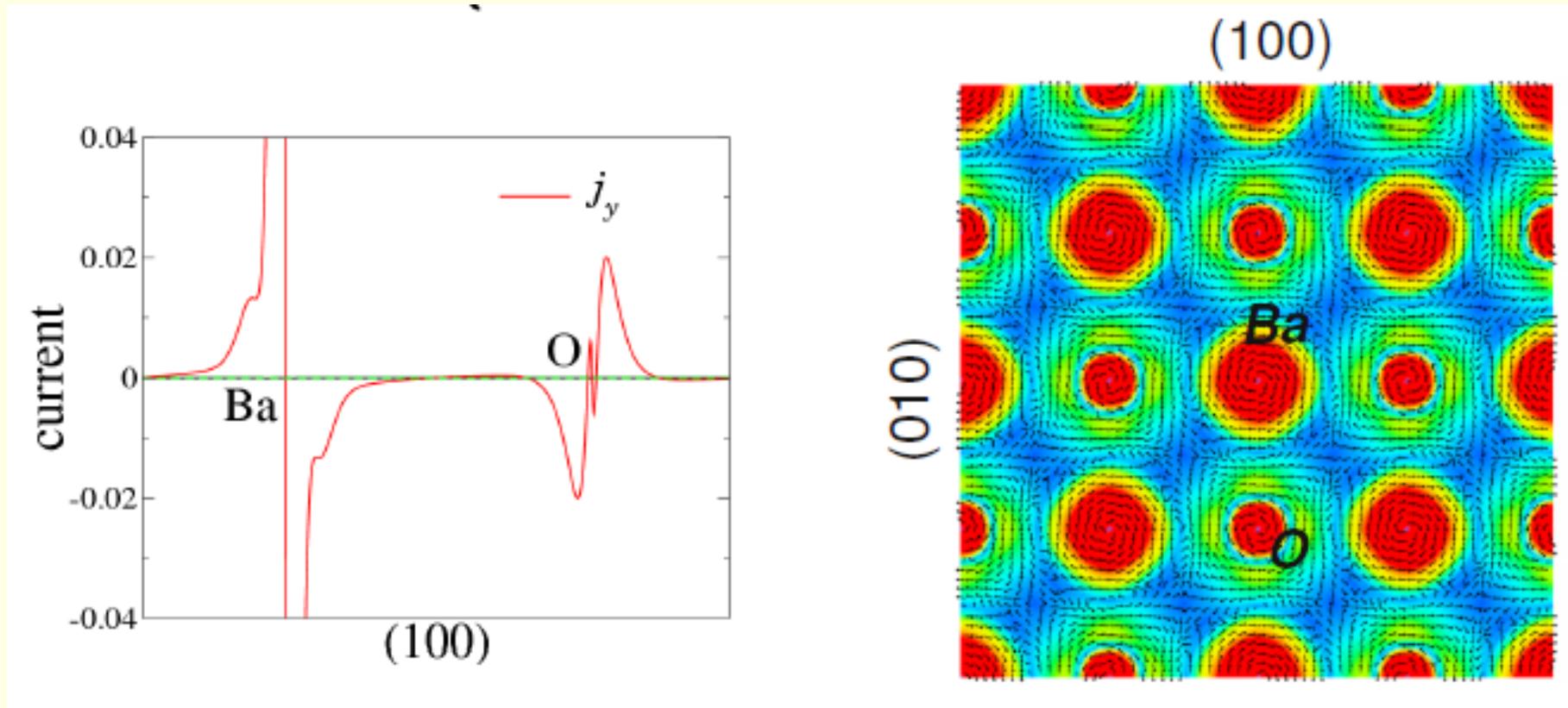
FIG. 1: Comparison of the induced current computed for an Ar atom with and without DUC and CC corrections (see text). For clarity only the valence 3s and 3p contribution is shown. j_ρ is the diamagnetic current calculated with the spherical charge density of 3s and 3p states using Eq.18.



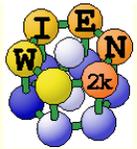


Induced current in LAPW

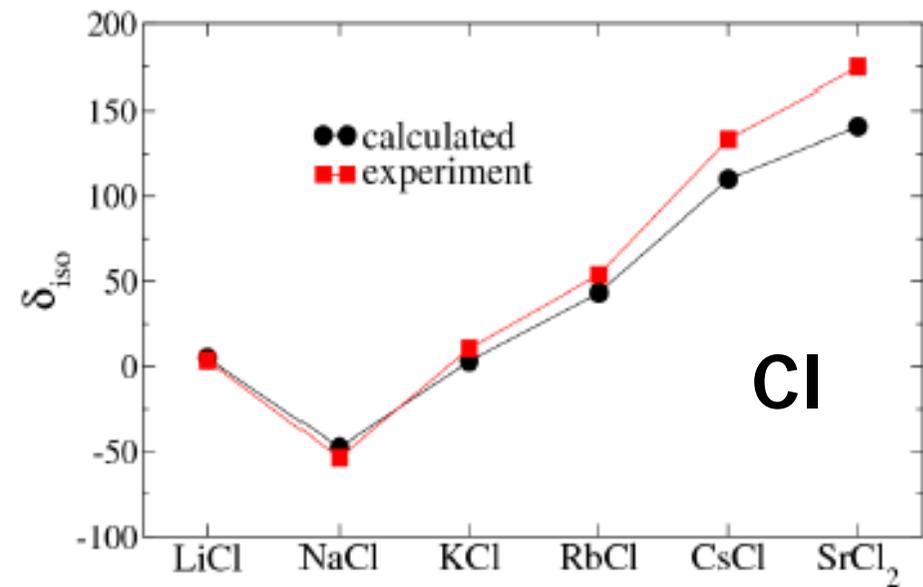
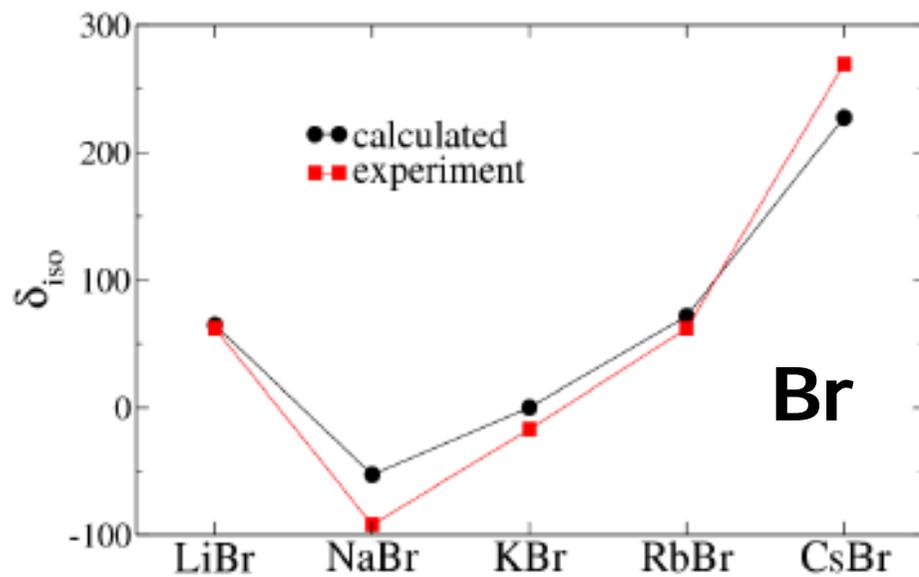
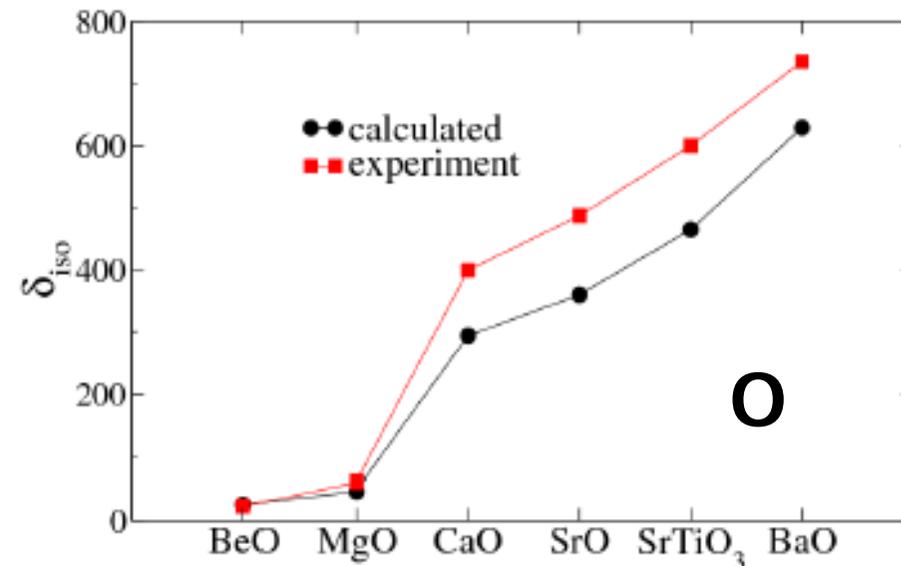
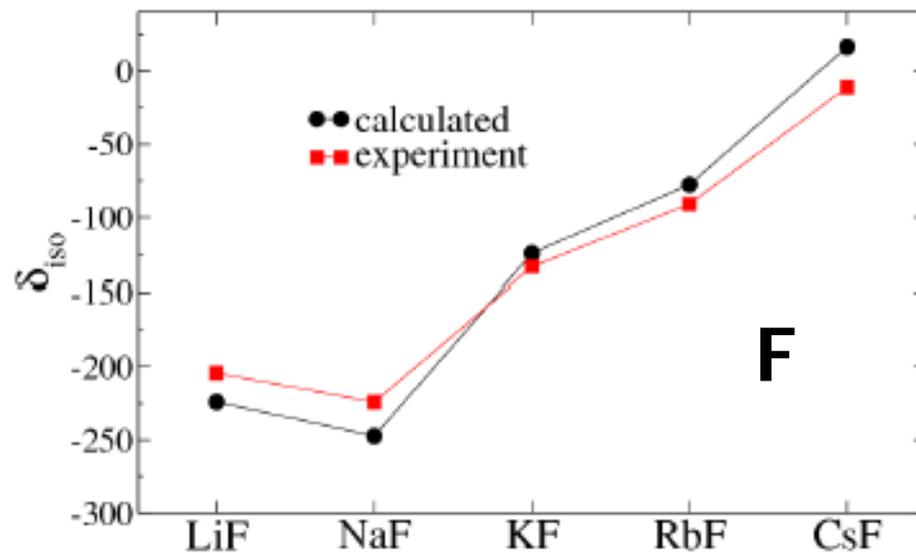
Induced current field for BaO (fcc) , B_{ext} in (001)



$$\mathbf{j}_{ind}(\mathbf{r}) = \begin{cases} \sum_{\mathbf{G}} \mathbf{j}_{\mathbf{G}} e^{i\mathbf{G}\cdot\mathbf{r}}, & \mathbf{r} \in I \\ \sum_{l,m} \mathbf{j}_{l,m}^{\alpha}(r) Y_{l,m}(\hat{r}). & \mathbf{r} \in S_{\alpha} \end{cases}$$



NMR shifts for F, O, Br, Cl





Interpretation of ^{19}F NMR shielding in alkali fluorides

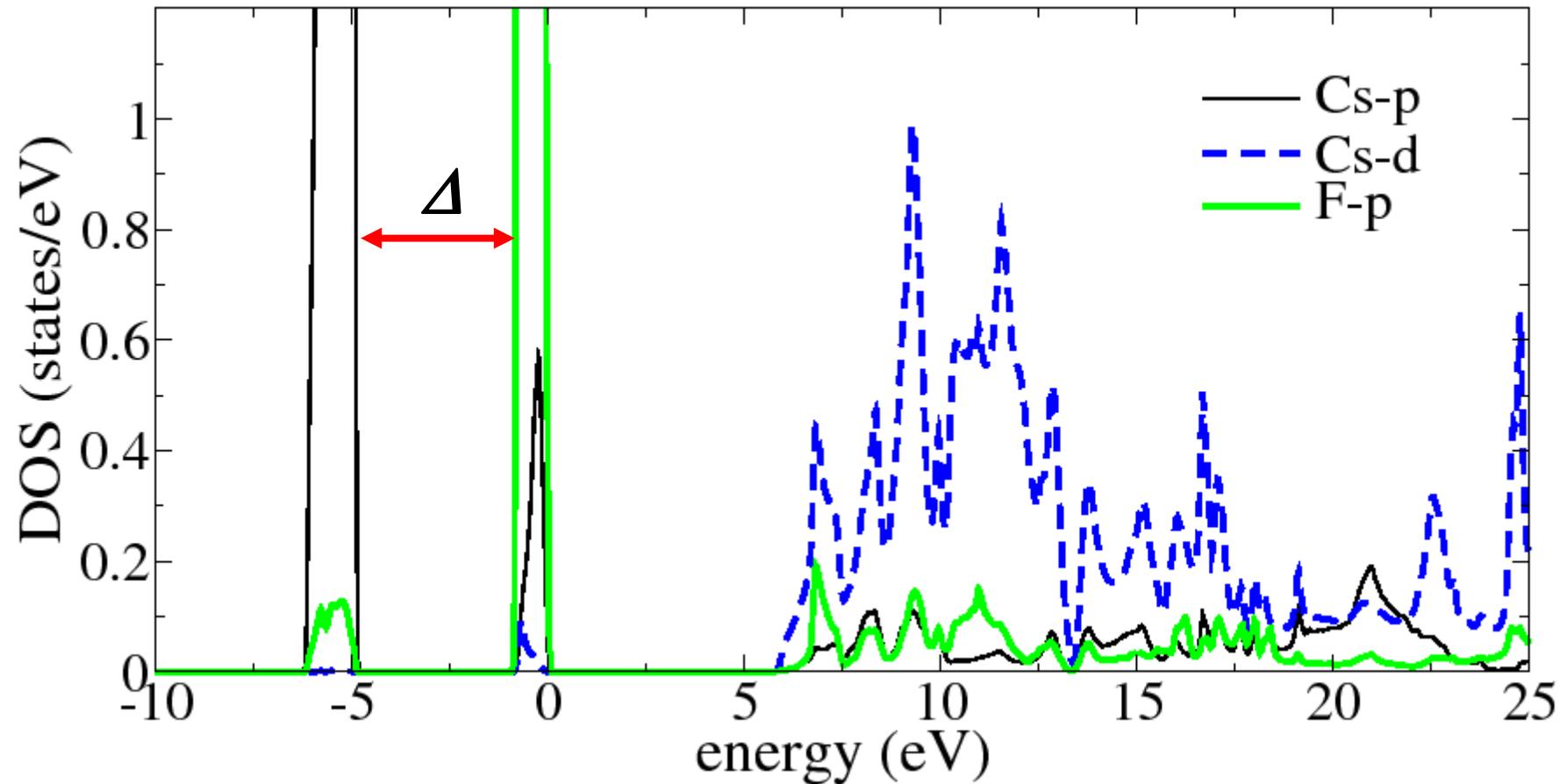


- band wise analysis
- character analysis (s,p,d) of the wave function of occupied and unoccupied states



DOS of alkali fluorides (CsF)

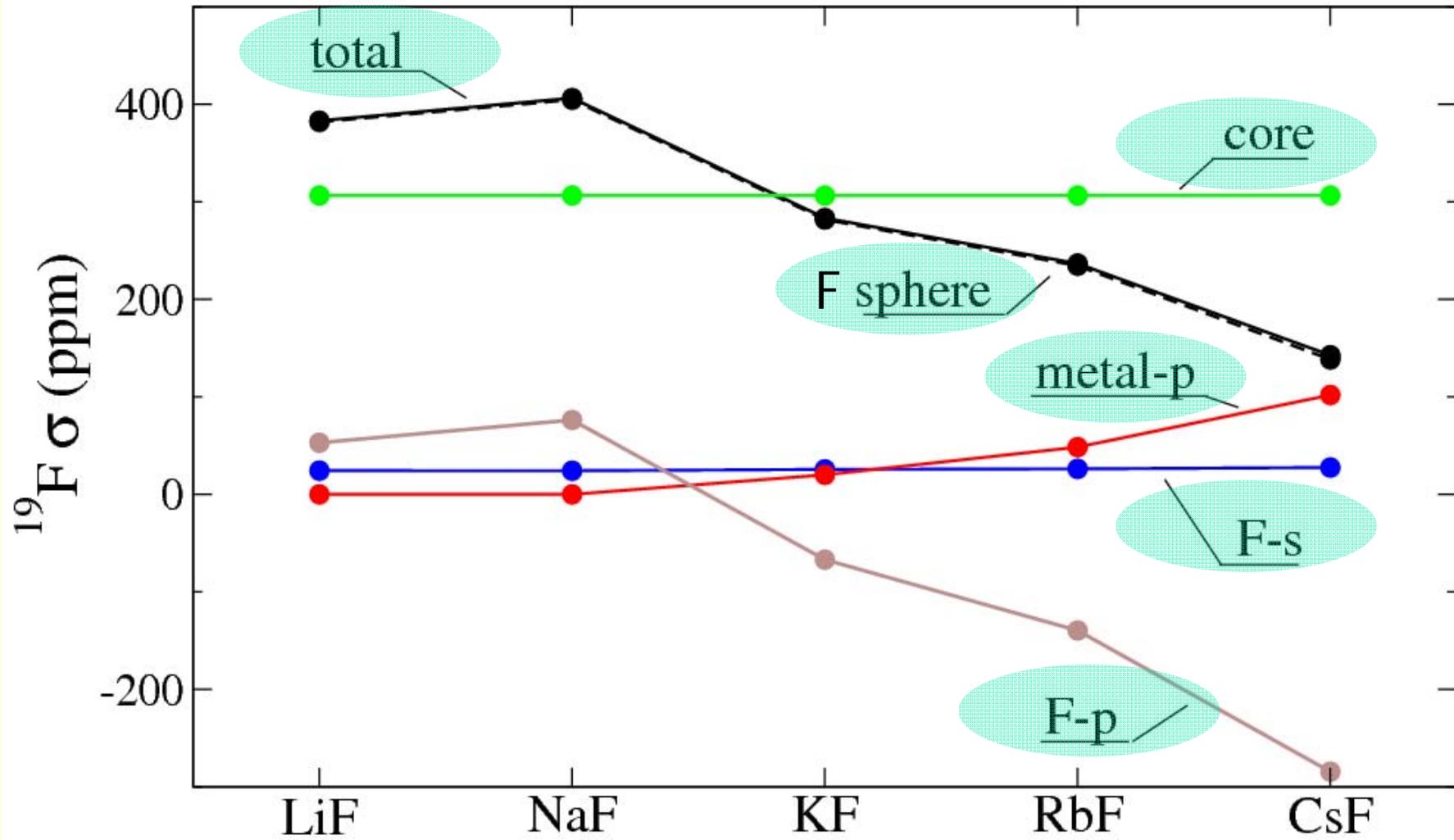
metal-p F-p band



Δ varies between 5 eV for CsF to 20 eV for NaF



Band wise analysis of the isotropic shielding in MF





Decomposition of NMR shift



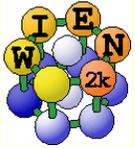
- decomposition of NMR shift according to s, p, d - character and atom

- $$\Psi_0 = \sum_{at} \sum_{lm} R_{at,lm} Y_{lm}$$

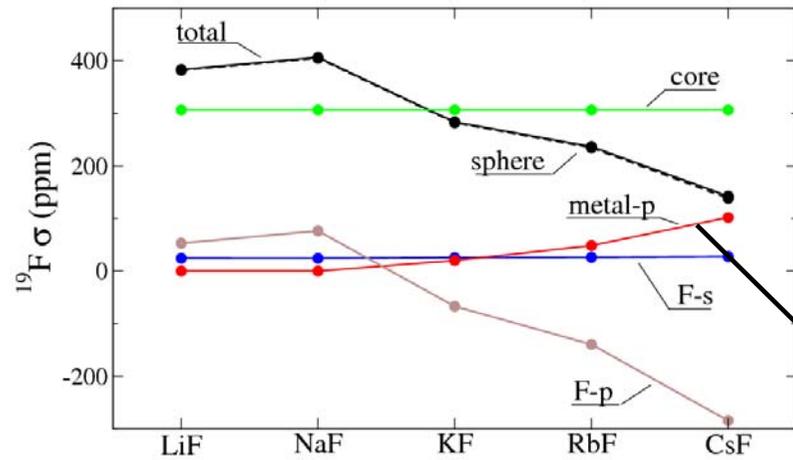
- decomposition according to ground state $\Psi_0^{(0)}$ and perturbed states $\Psi_0^{(1)}$

$$\mathbf{j}_{ind}(\mathbf{r}') = \frac{1}{c} \sum_o \Re \left[\langle \Psi_0^{(0)} | \mathbf{J}^p(\mathbf{r}') G(\epsilon_o)(\mathbf{r}-\mathbf{r}') \times \mathbf{p} \cdot \mathbf{B} | \Psi_0^{(0)} \rangle \right]$$

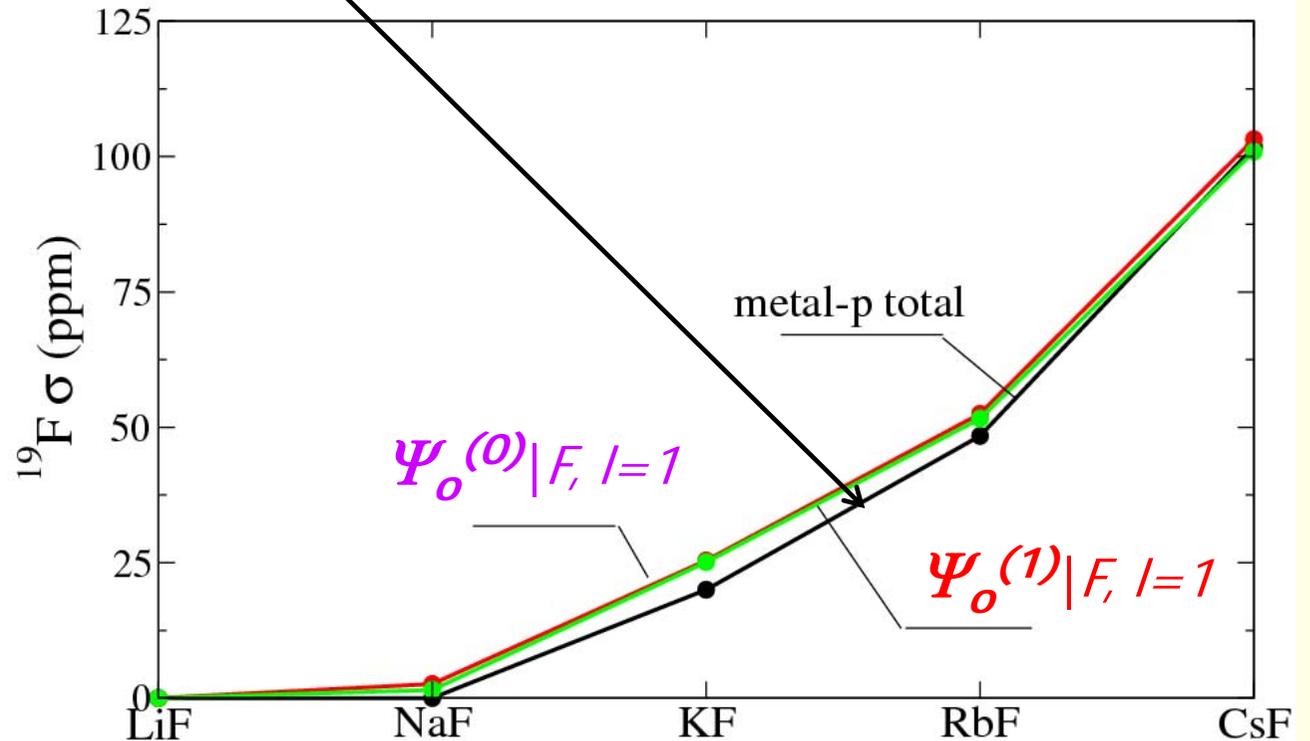
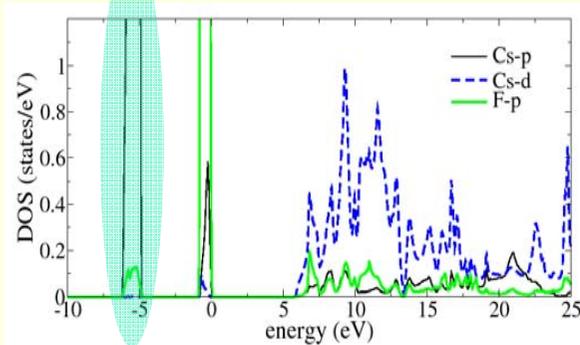
$\Psi_0^{(1)}$



metal-p band contribution

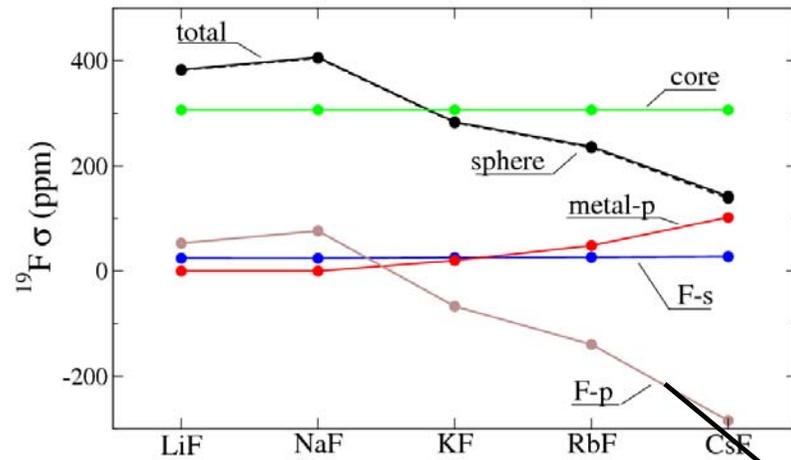


$$j_{ind}(\mathbf{r}') = \frac{1}{c} \sum_o \Re [\langle \Psi_o^{(0)} | \mathbf{J}^P(\mathbf{r}') | \Psi_o^{(1)} \rangle]$$

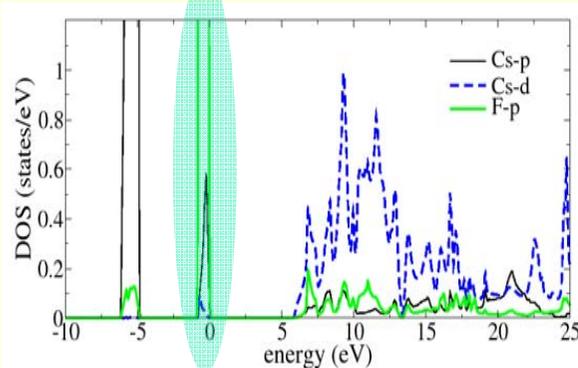
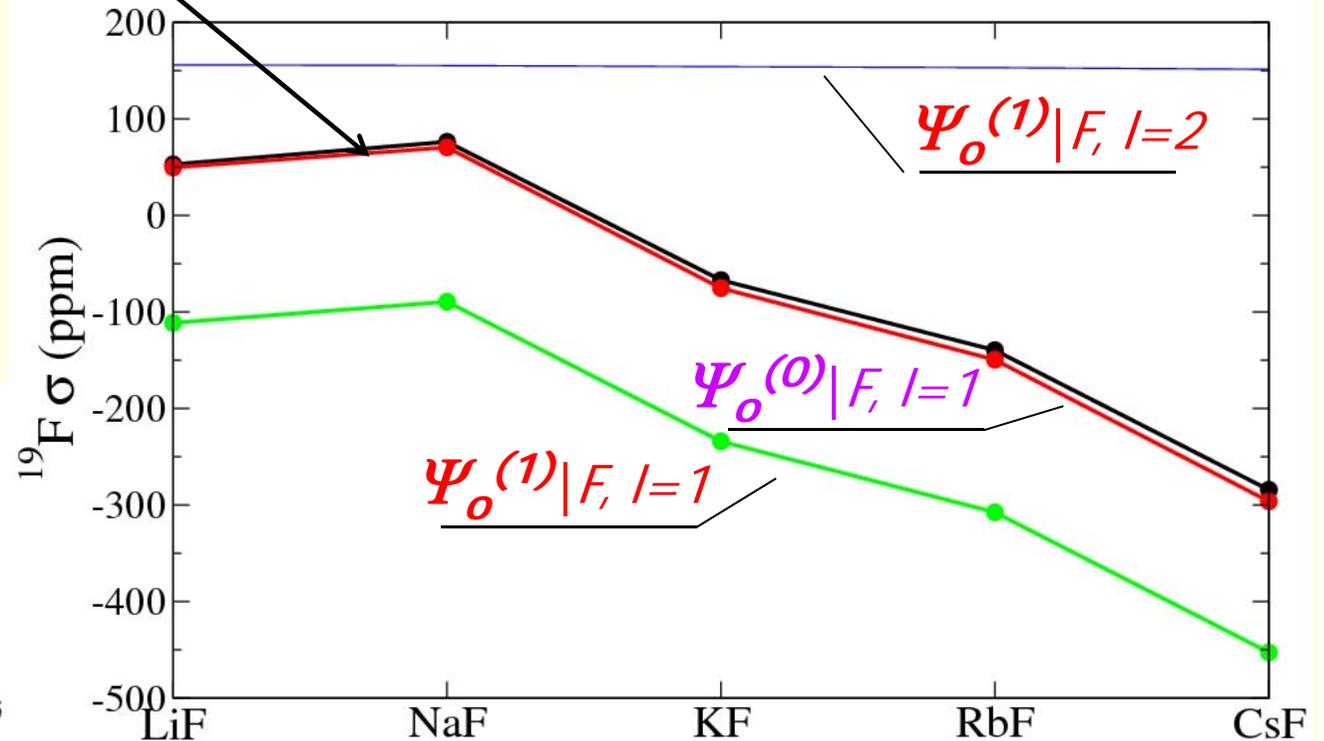


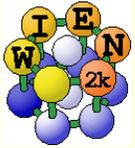


F-p band contribution

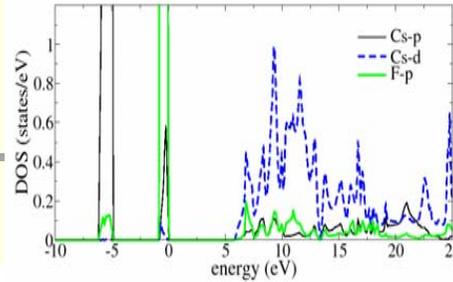


$$j_{ind}(\mathbf{r}') = \frac{1}{c} \sum_o \Re [\langle \Psi_o^{(0)} | \mathbf{J}^P(\mathbf{r}') | \Psi_o^{(1)} \rangle]$$

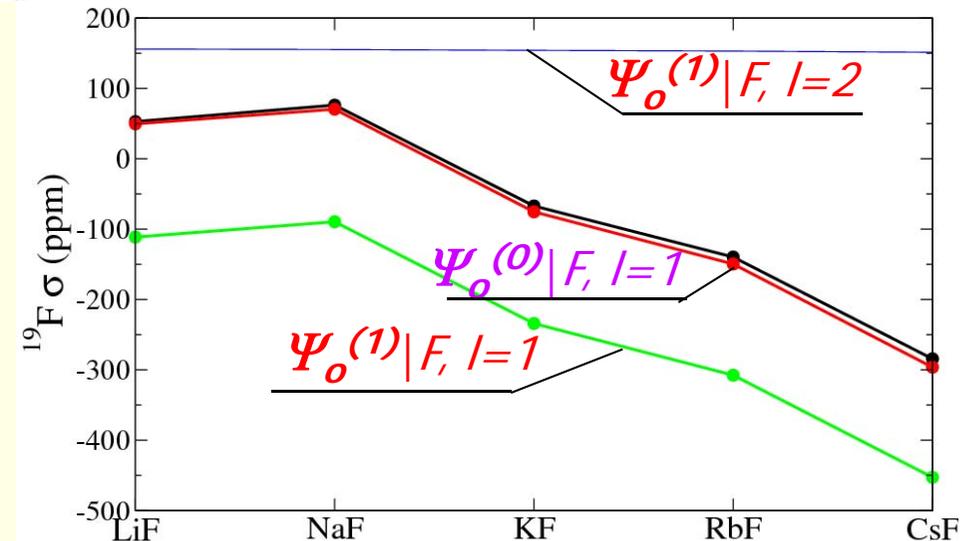
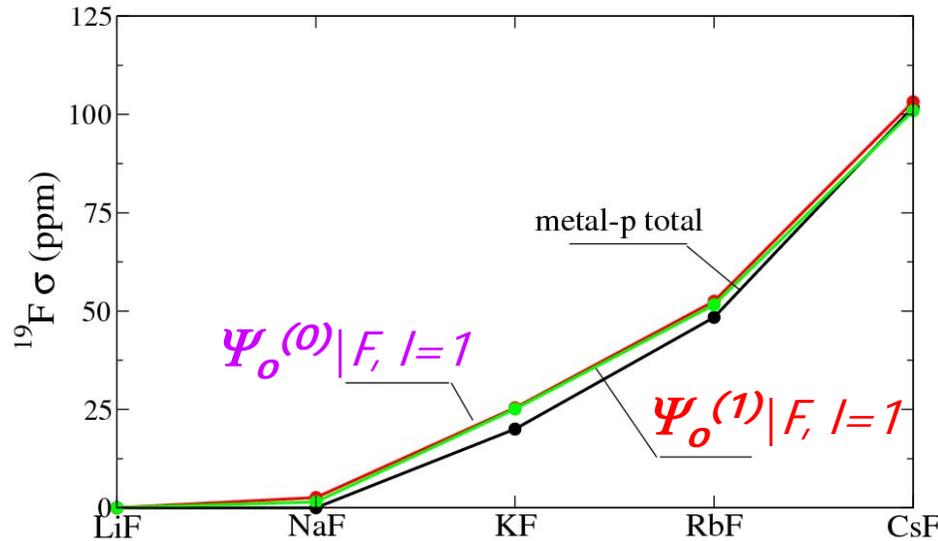




metal p band



F-p band



the only important ground state contribution

$$\Psi_0^{(0)} | F, l=1$$

the only important ground state contribution

positive, increasing contribution within the series

$$\Psi_0^{(1)} | F, l=1$$

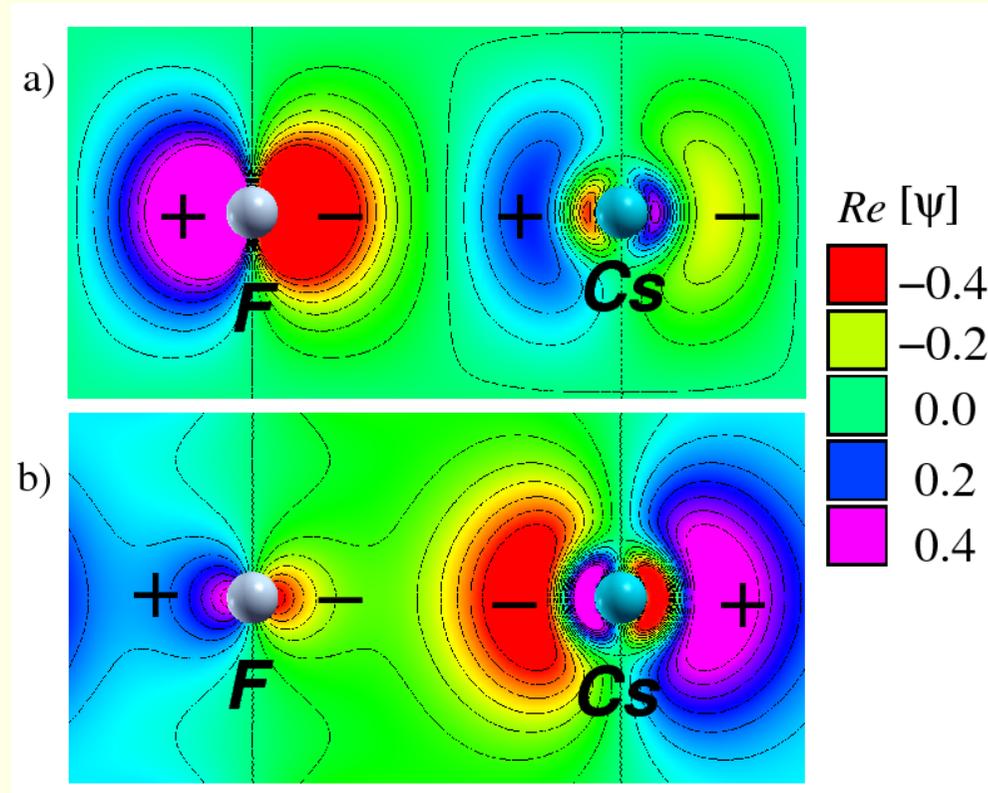
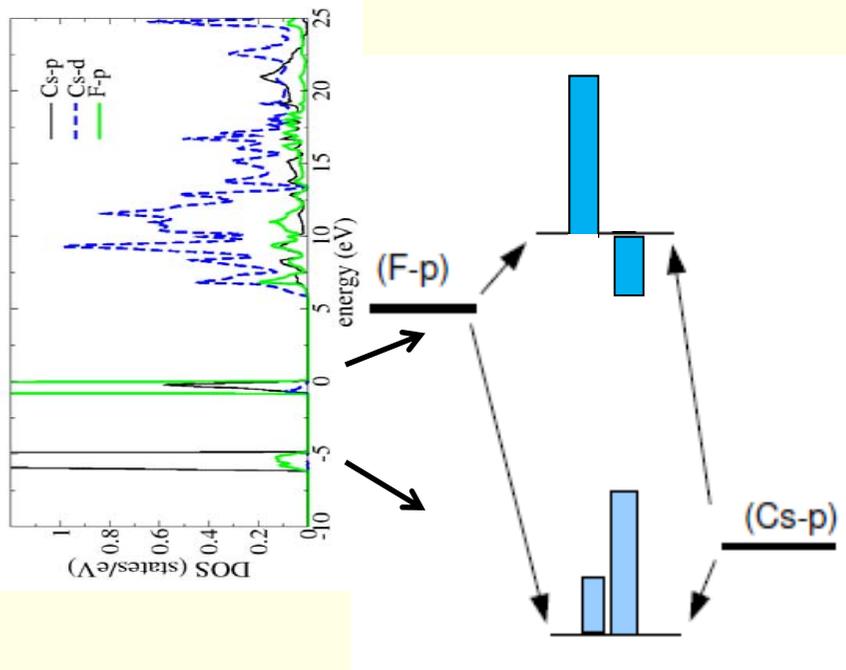
negative, decreasing contribution within the series

$$\Psi_0^{(1)} | F, l=2$$

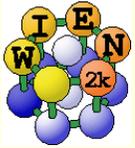
constant contribution within the series



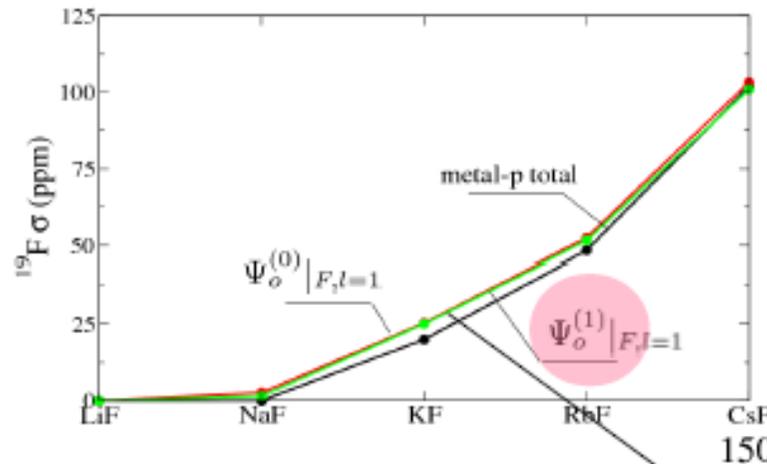
Re[Ψ] at X-point of CsF



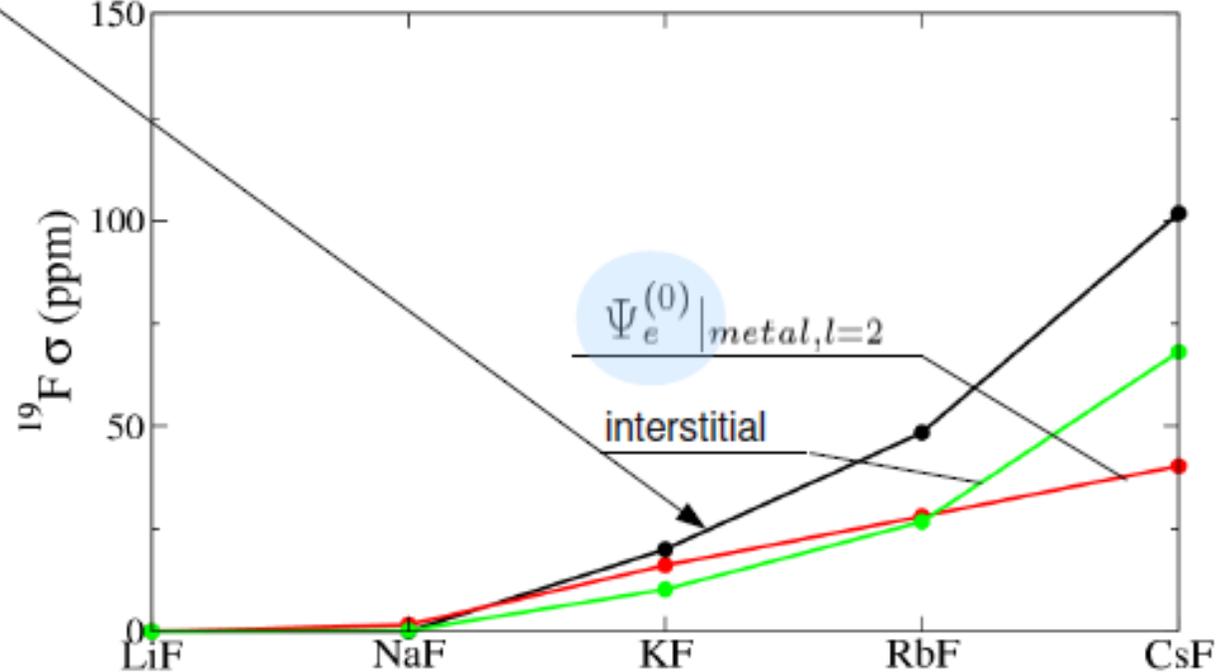
- a) F-p band, **anti-bonding** character of the Cs-p and F-p orbitals, **negative** contribution to the shielding
- b) Cs-p band, **bonding** character between Cs-p and F-p orbitals, **positive** contribution to the shielding.

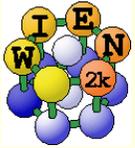


metal-p band contribution

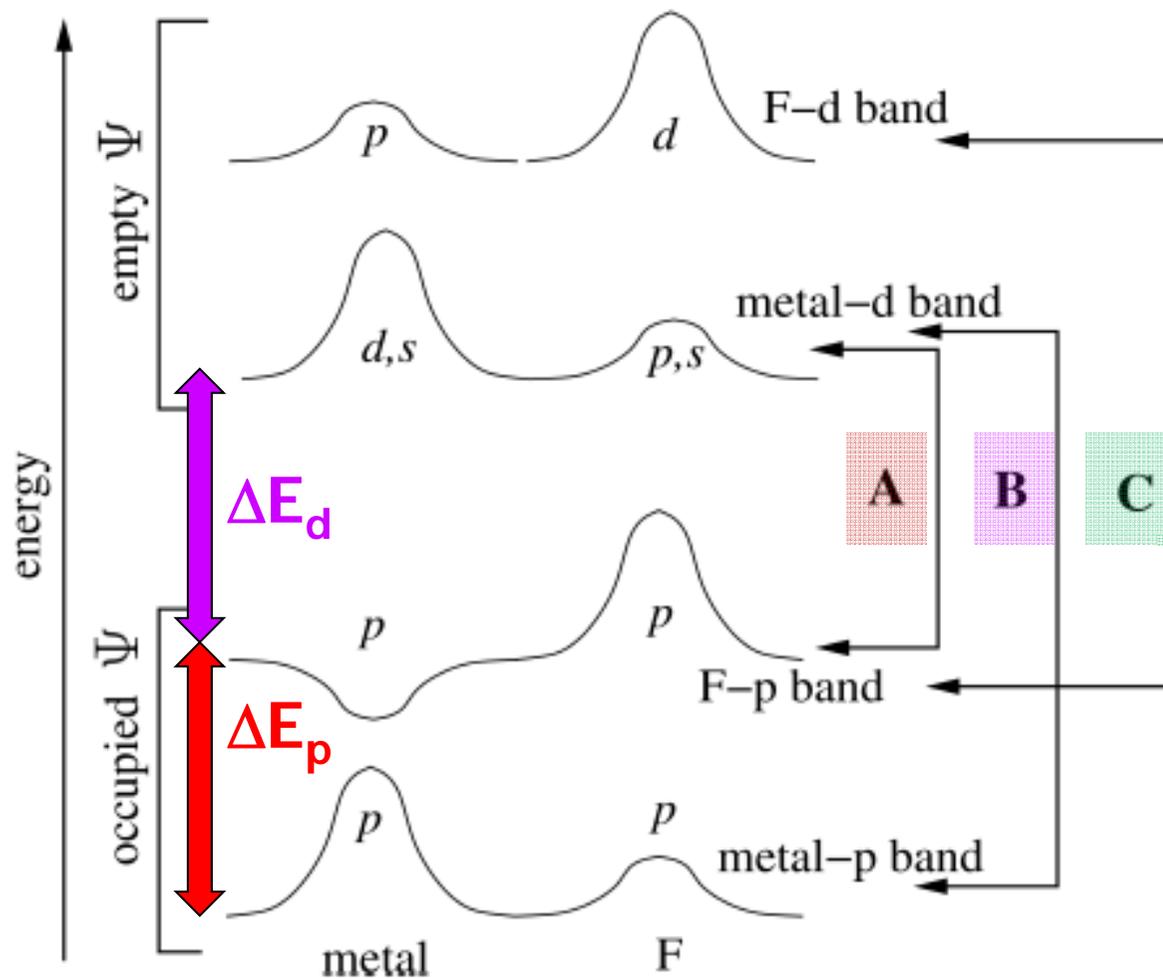


$$|\Psi_o^{(1)}\rangle = \sum_e |\Psi_e^{(0)}\rangle \frac{\langle \Psi_e^{(0)} | H^{(1)} | \Psi_o^{(0)} \rangle}{\epsilon - \epsilon_e}$$





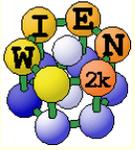
Interactions relevant for NMR chemical shifts in alkali fluorides



A, contributes to $\Psi_o^{(1)}|_{F,l=1}$ component, negative contribution to the shielding.

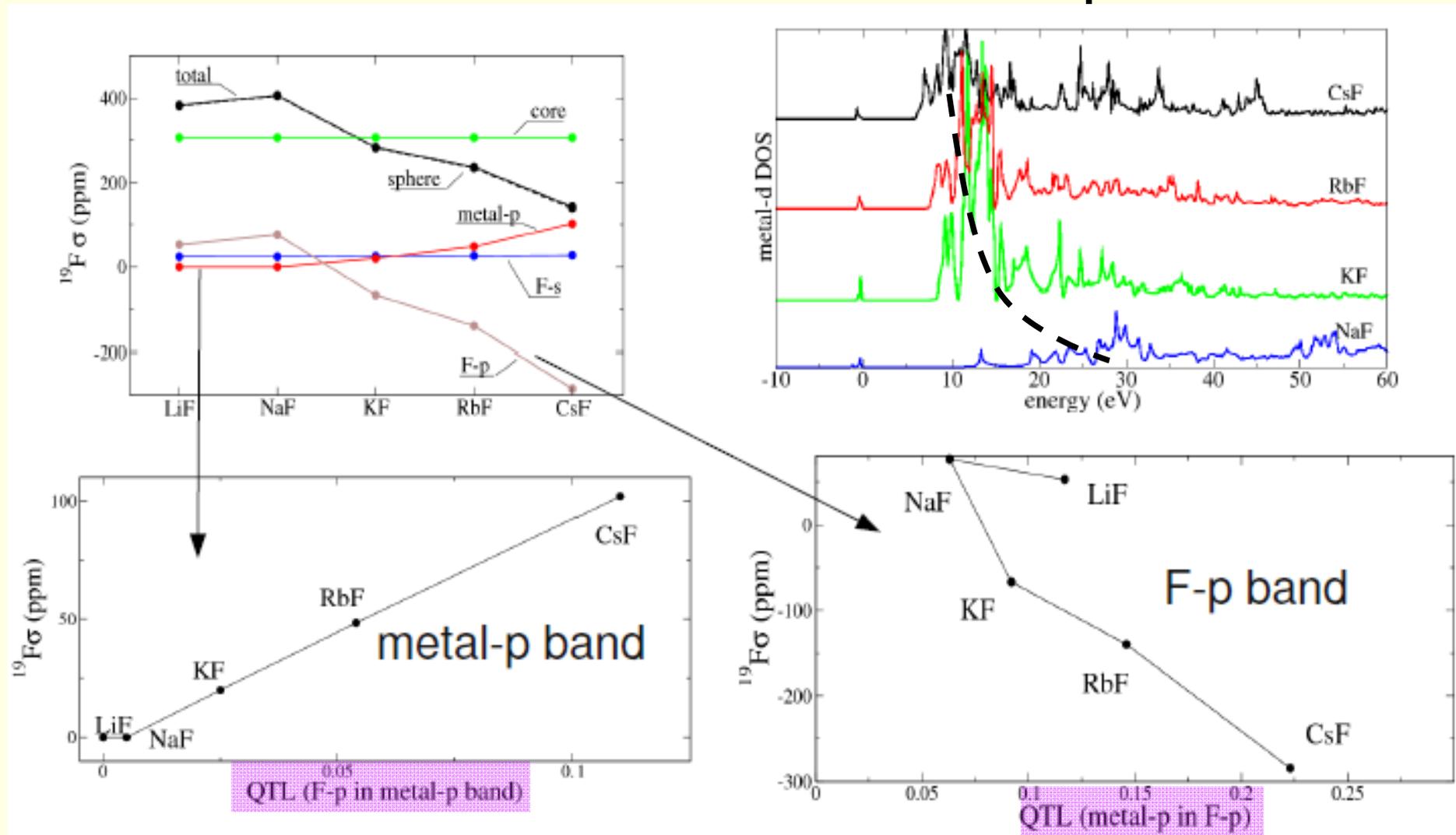
B, contributes to $\Psi_o^{(1)}|_{F,l=1}$ component, positive contribution to the shielding.

C, contributes to $\Psi_o^{(1)}|_{F,l=2}$ constant contribution to the shielding.



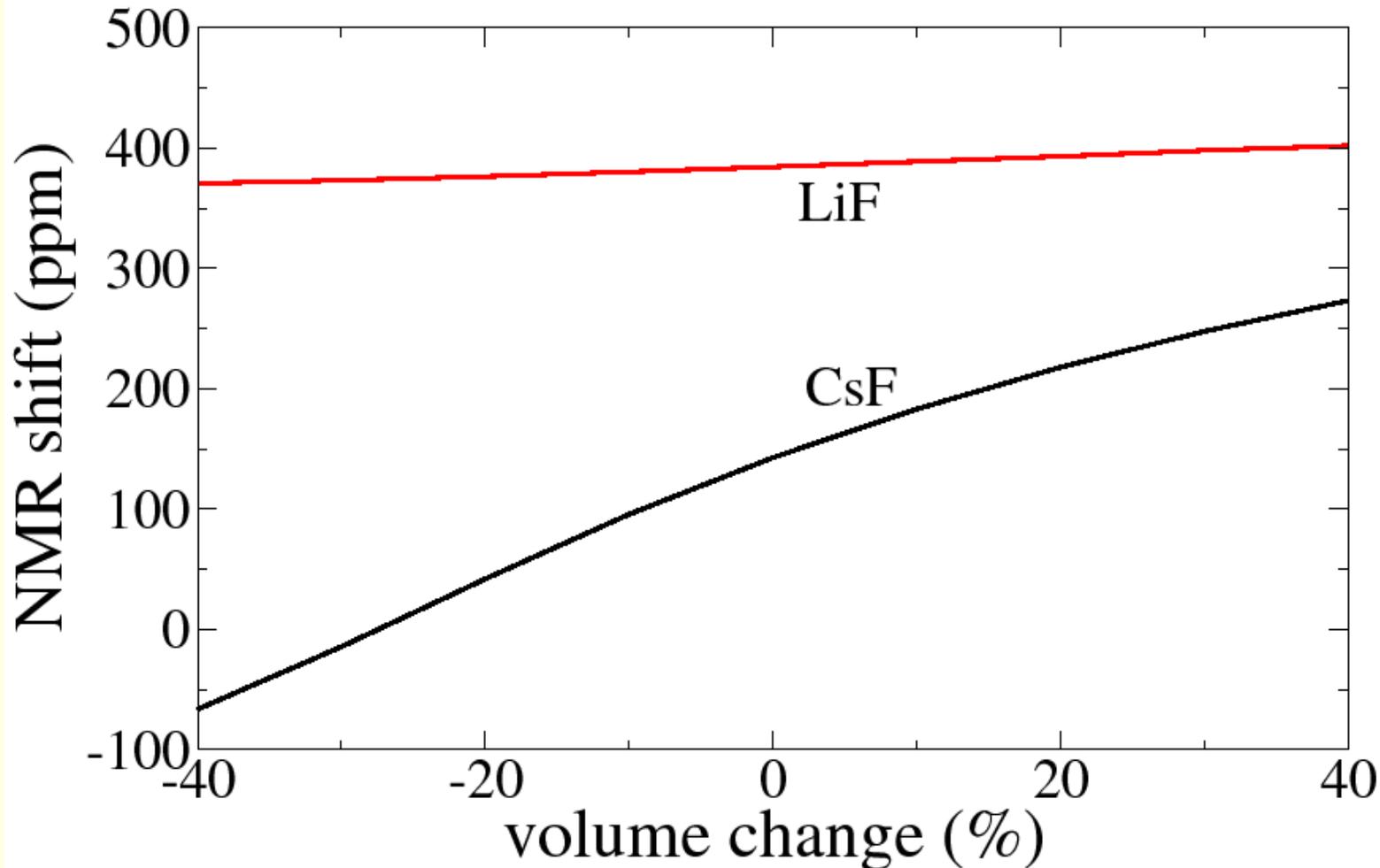
A, B interactions:

coupling to the metal-d states, due to F-p – metal-p hybridization
d-band position





Effect of bond distance on the shielding



- decreasing volume leads to stronger Me-p F-p interaction and to more negative shielding (Li does not have "Li-p band")

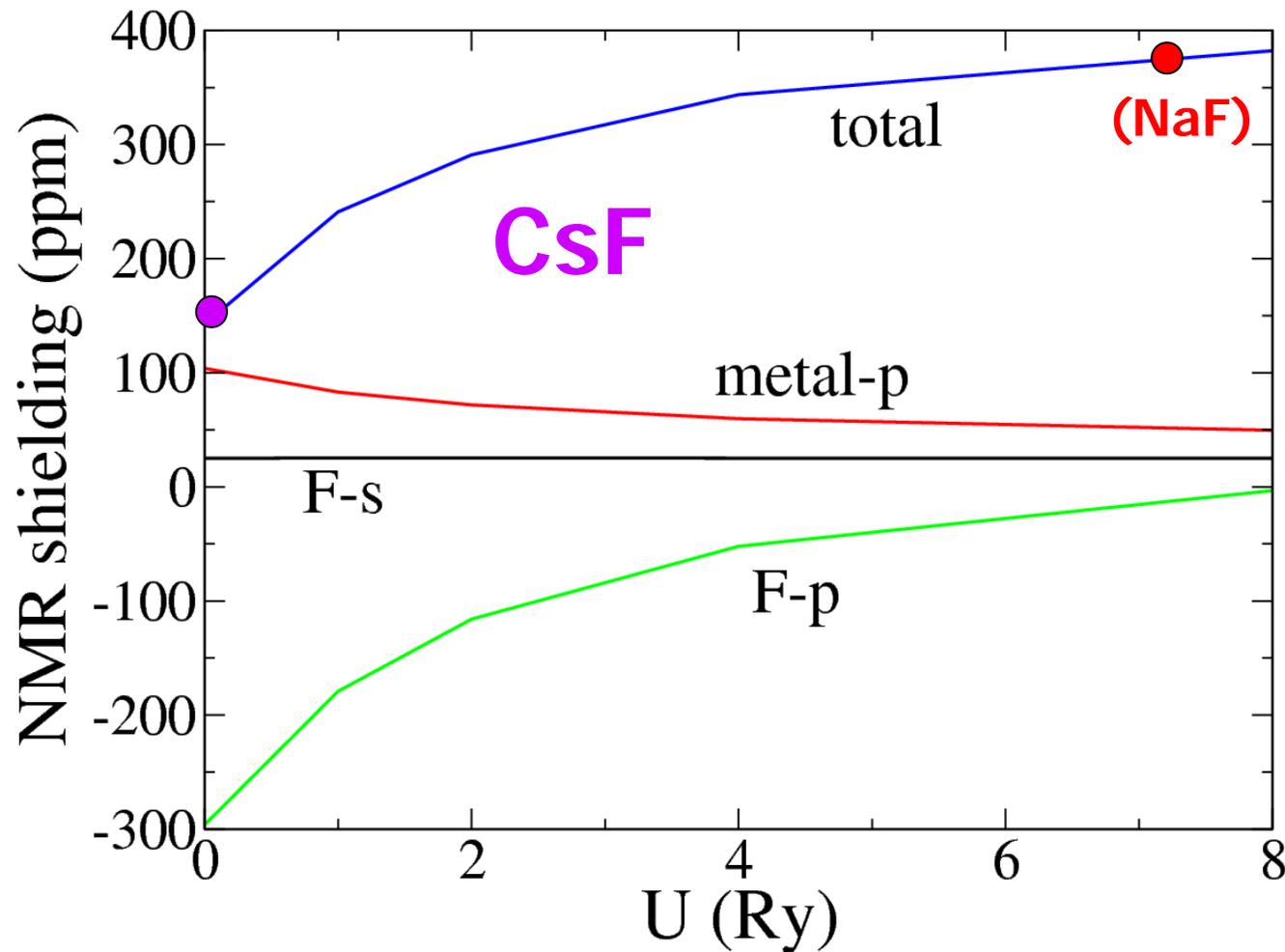


Effect of position of metal-d band on the F shielding



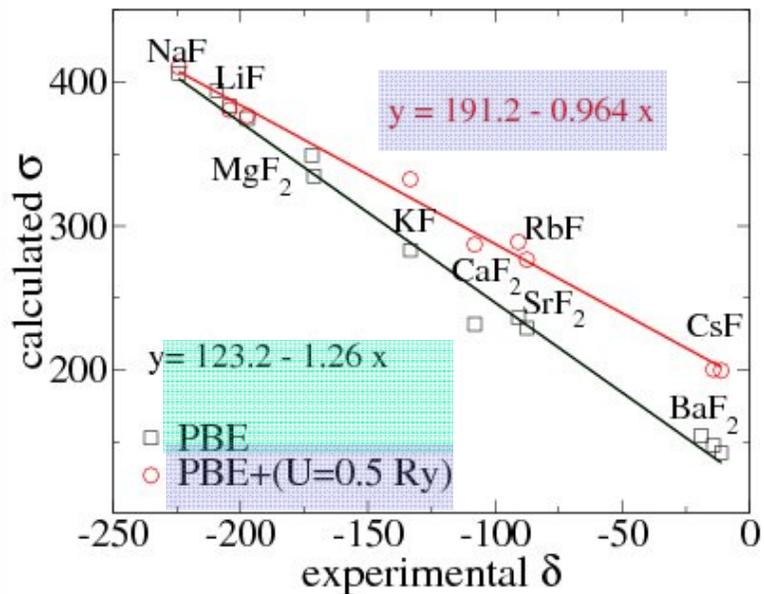
LDA+U acting on Cs-d

$$|\Psi_o^{(1)}\rangle = \sum_e |\Psi_e^{(0)}\rangle \frac{\langle \Psi_e^{(0)} | H^{(1)} | \Psi_o^{(0)} \rangle}{\epsilon - \epsilon_e}$$





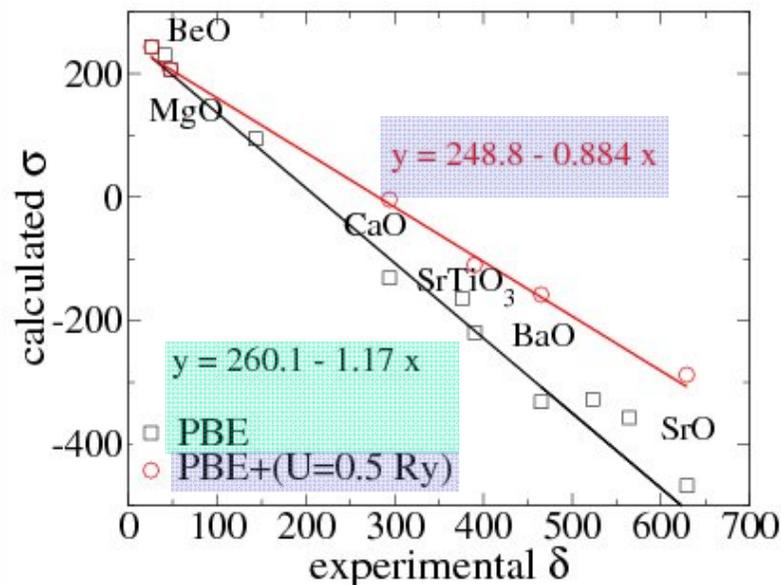
The "slope" - problem



exp. δ vs. theoretical σ :
The slope must be **ONE**

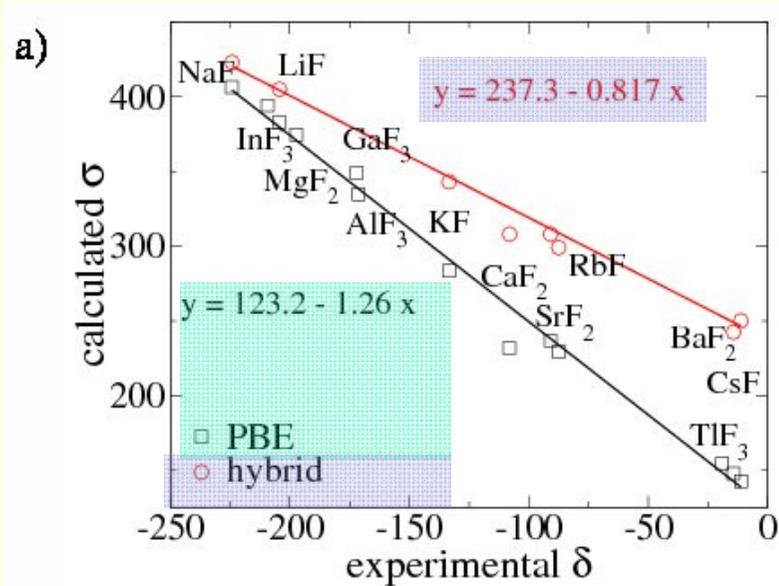
PBE: slope is too big

PBE+U (metal d-states):
with **one** U value it is **not**
possible to fix **oxygen AND**
fluorine CS.



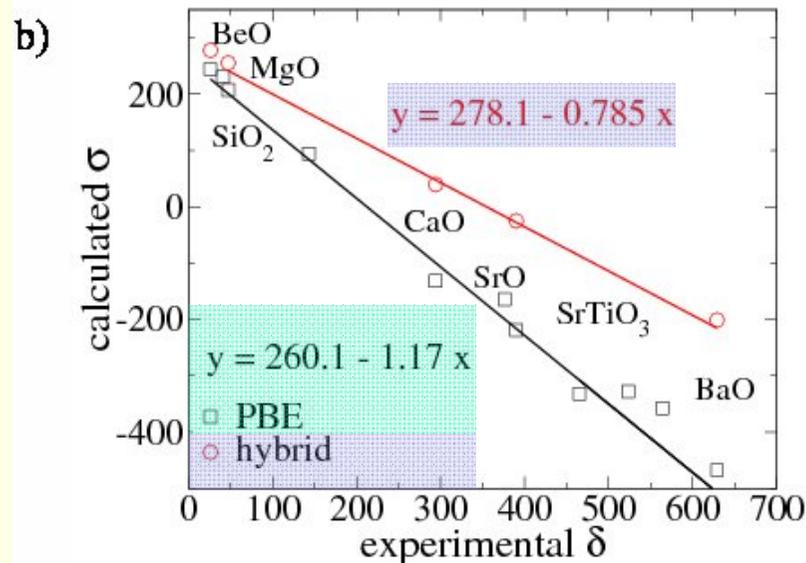


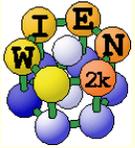
the slope - problem



- hybrid-DFT is the standard method in CS calculations of molecules (Gaussian)

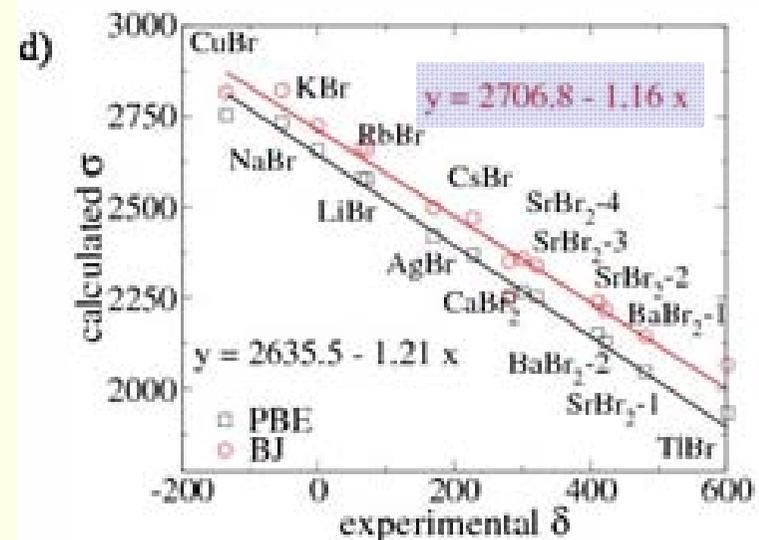
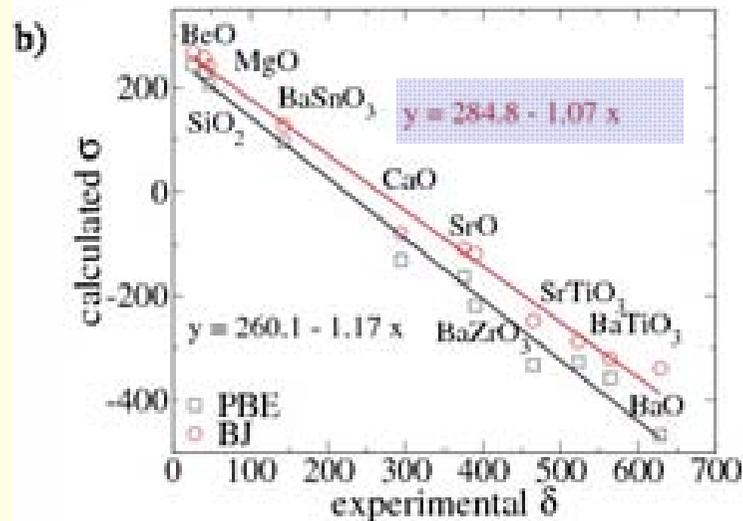
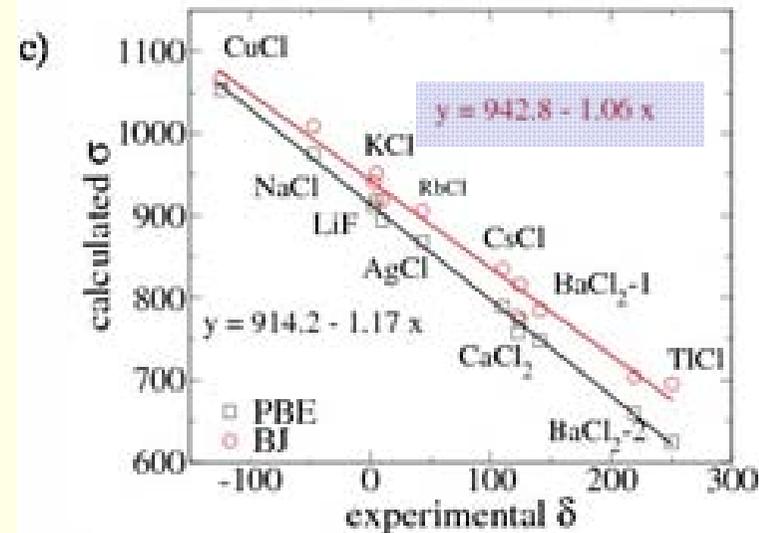
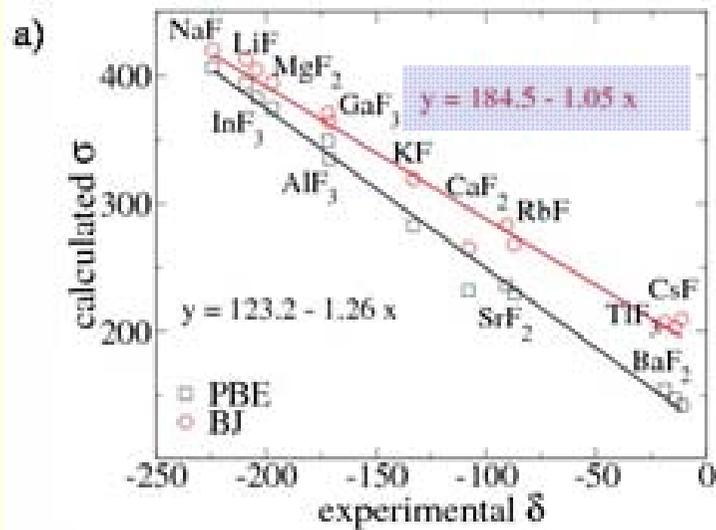
- for (ionic) solids YS-PBE0 (HSE) gives a much **too large** correction (smaller mixing ??)





the slope - problem

BJ-potential (OEP) seems quite reasonable for ionic compounds



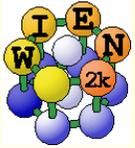


Summary:



NMR chemical shifts:

- shielding of anions in solids determined by:
 - *strength of metal-p -- F-p hybridization*
 - distance of metal-p band from anion-p band
 - bond distance, number of neighbors
 - *position of empty metal-d states*



Acknowledgement



Robert Laskowski (TU Vienna)

NMR: PRB **85**, 035132 (2012)
PRB **85**, 245117 (2012)



Thank you for your attention !