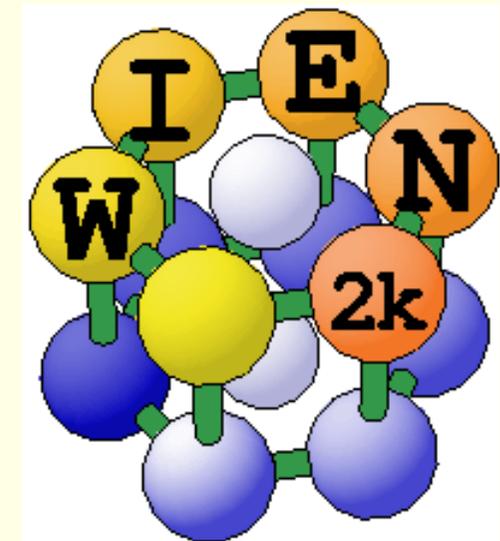


# Calculations of NMR chemical shifts in solids

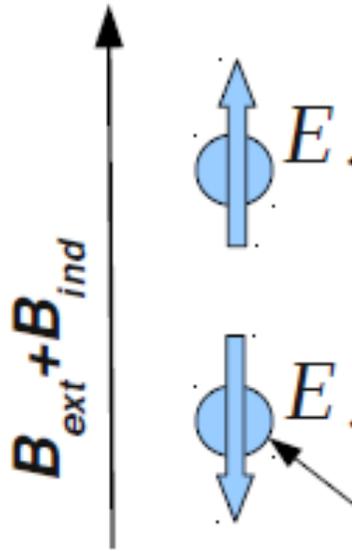
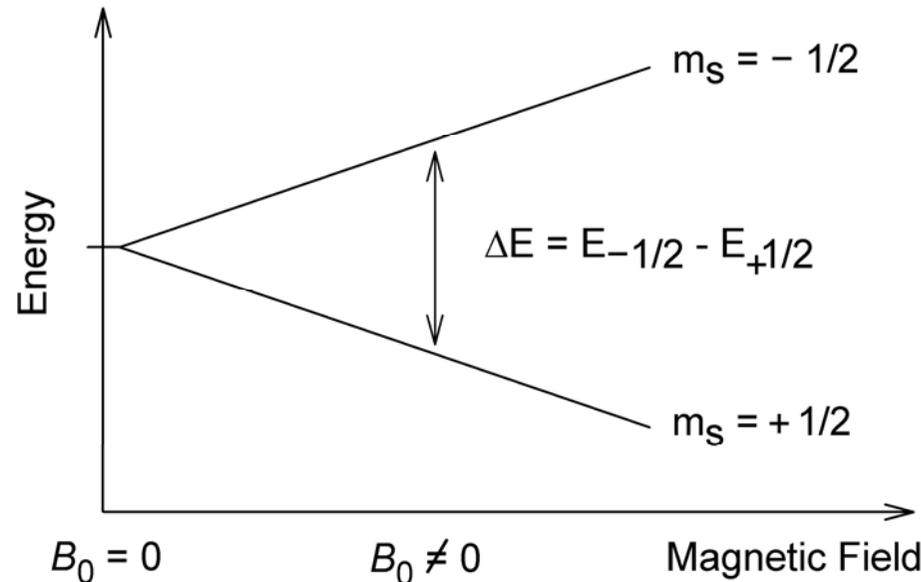
**Peter Blaha**

Institute of Materials Chemistry  
TU Vienna, Austria





# NMR spectroscopy



$$\Delta E = -\gamma m \hbar (B_{ext} + B_{ind})$$

nucleus with dipole moment

$B_{ext} + B_{ind}$  is measured at any nucleus by detecting transition energy related to reorientation of its magnetic moment



# NMR Hamiltonian



perturbation

$$H_{NMR} = H_Z + H_\sigma + H_Q + H_D + H_J + \dots$$

$$H_Z = -\mu \cdot B_{ext}$$

Zeeman Hamiltonian

electric quadrupole  
interaction (EFG)

Indirect spin-  
spin coupling

direct dipolar  
coupling

$$H_\sigma = -\mu \cdot B_{ind}$$

magnetic shielding



# NMR Hamiltonian



$$H_{NMR} = H_Z + H_\sigma + H_Q + H_D + H_J + \dots$$

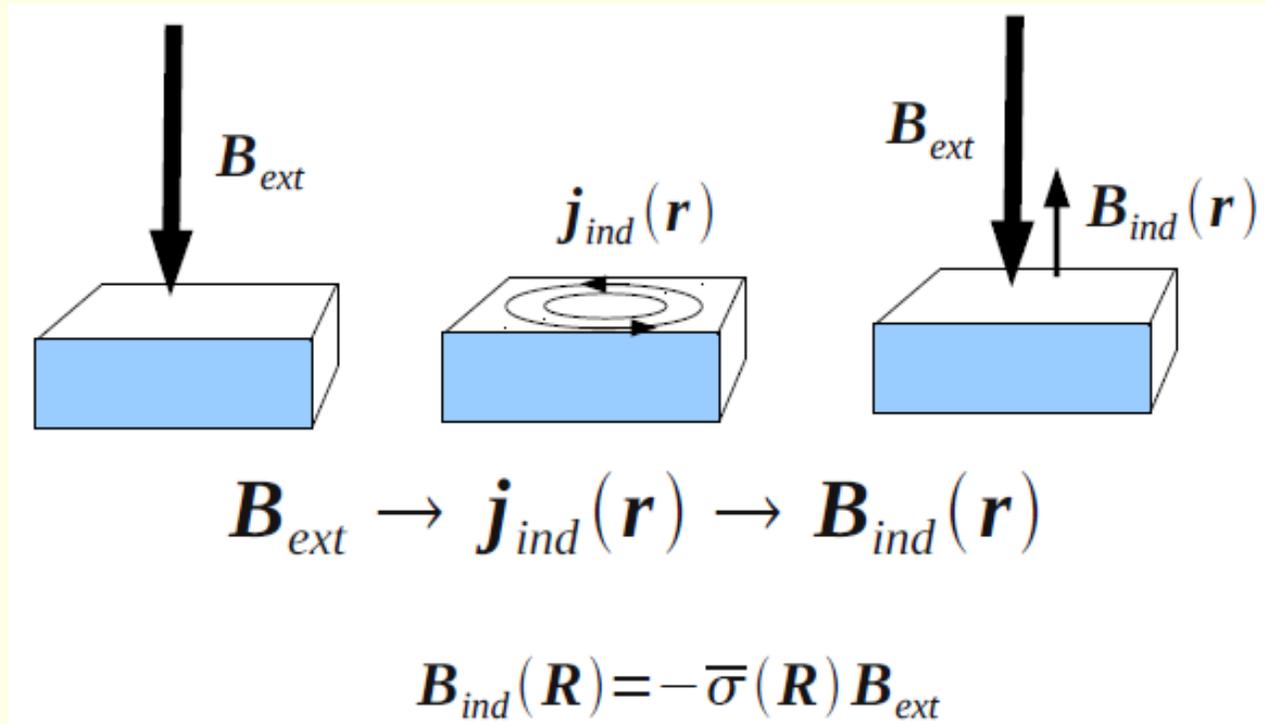
$H_Z = -\mu \cdot B_{ext}$   
Zeeman Hamiltonian

$H_\sigma = -\mu \cdot B_{ind}$   
magnetic shielding

$H_Q \approx eQ\Phi / h$   
quadrupole interaction



# NMR shielding, chemical shift:



$\sigma(\mathbf{R})$  is the **shielding tensor** at the nucleus  $\mathbf{R}$

**chemical shift:**

$$\delta(\text{ppm}) = \frac{\sigma_{ref} - \sigma}{1 - \sigma_{ref}} \times 10^6$$



# Biot - Savart law



The induced **magnetic field** ( $\mathbf{B}_{ind}$ ) is derived from the **induced current** ( $\mathbf{j}_{ind}$ ) using a standard formula:

$$\mathbf{B}_{ind}(\mathbf{r}) = \frac{1}{c} \int d^3 r' \mathbf{j}(\mathbf{r}') \times \frac{\mathbf{r} - \mathbf{r}'}{|\mathbf{r} - \mathbf{r}'|^3}$$

in DFT the **current density**  $\mathbf{j}(r)$  will be:

$$\mathbf{j}_{ind}(\mathbf{r}') = \sum_o \left[ \langle \Psi_o^{(1)} | \mathbf{J}^{(0)}(\mathbf{r}') | \Psi_o^{(0)} \rangle + \langle \Psi_o^{(0)} | \mathbf{J}^{(0)}(\mathbf{r}') | \Psi_o^{(1)} \rangle \text{ diamagn.} \right. \\ \left. + \langle \Psi_o^{(0)} | \mathbf{J}^{(1)}(\mathbf{r}') | \Psi_o^{(0)} \rangle \right], \quad (4) \text{ paramagn.}$$

**perturbed w.f.**  $\Psi^1$  is obtained from perturbation theory

$$|\Psi_o^{(1)}\rangle = \sum_e |\Psi_e^{(0)}\rangle \frac{\langle \Psi_e^{(0)} | H^{(1)} | \Psi_o^{(0)} \rangle}{\epsilon_o - \epsilon_e}$$

sum over **all** empty states

$$H^{(1)} = \frac{1}{2c} \mathbf{r} \times \mathbf{p} \cdot \mathbf{B}$$

magnetic field



sum over **ALL** empty states:



$$|\Psi_o^{(1)}\rangle = \sum_e |\Psi_e^{(0)}\rangle \frac{\langle \Psi_e^{(0)} | H^{(1)} | \Psi_o^{(0)} \rangle}{\epsilon_o - \epsilon_e}$$

- standard APW basis set  $u_l(r, E_l)$  only good near linearization  $E_l$
- adding additional LOs at high energies (up to 1000 Ry !!!)
- $H^{(1)}$  contains the  $\nabla$  operator, so we need to represent the radial derivative of  $u_l(r, E_l)$  at  $l \pm 1$

- adding "NMR-los"  $\xi_{l,k}(r, \tilde{\epsilon}) = \begin{cases} r \frac{d}{dr} u_{l+1}(r, \tilde{\epsilon}) + (l+2)u_{l+1}(r, \tilde{\epsilon}), & k=1 \\ r \frac{d}{dr} u_{l-1}(r, \tilde{\epsilon}) - (l-1)u_{l-1}(r, \tilde{\epsilon}), & k=2 \end{cases}$

- x\_nmr -mode in1 [-focus nmr\_atom] will set that up automatically



# practical calculation



- run normal scf cycle
- `x_nmr_lapw -mode in1 [-focus 0]`
  - view the resulting `*in1c_nmr` file
- `x_nmr_lapw [-p]`

- creates several directories (`nmr_q0`, `nmr_pqx`, `nmr_mqx`, `nmr_pqy`, ..) and performs `lapw1/2` steps for several  $k$ -meshes ( $k \pm q$ )
- creates the current
- integrates the current

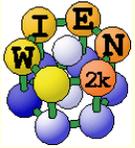
$$\mathbf{r} \cdot \hat{\mathbf{u}}_i = \lim_{q \rightarrow 0} \frac{1}{2q} \left( e^{iq\hat{\mathbf{u}}_i \cdot \mathbf{r}} - e^{-iq\hat{\mathbf{u}}_i \cdot \mathbf{r}} \right)$$

- `tail case.outputnmr_integ`

- `:NMRTOT003 ATOM: O 3 NMR (total/ppm) TRACE = -366.9034 ANISO = -264.8134 ASYM = -0.9504`

- for analysis one can calculate the shift from certain bands (energy range) only
  - `x_nm_lapw [-p] -noinit -emin xx [-emax yy]`

```
WFFIL EF=.533144859350 (WFFIL, WFPRI, ENFIL,
7.00 10 4 (R-MT*K-MAX; MAX L IN WF, V-NM
0.30 19 0 (GLOBAL E-PARAMETER WITH n OTH
0 -0.58576 0.002 CONT 1
0 4.80000 0.000 CONT 1
0 36.60000 0.000 CONT 1
0 66.66000 0.000 CONT 1
0 104.26000 0.000 CONT 1
0 149.26000 0.000 CONT 1
0 201.50000 0.000 CONT 1
```



# Test of accuracy: Ar atom



- the current  $\mathbf{j}$  and chemical shielding  $\sigma$  of a **spherical atom** can be calculated “exactly” from the density  $\rho(\mathbf{r})$  (no perturbation theory) by:

$$\mathbf{j}_\rho(\mathbf{r}') = -\frac{1}{2c}\rho(\mathbf{r}')\mathbf{B} \times \mathbf{r}'$$

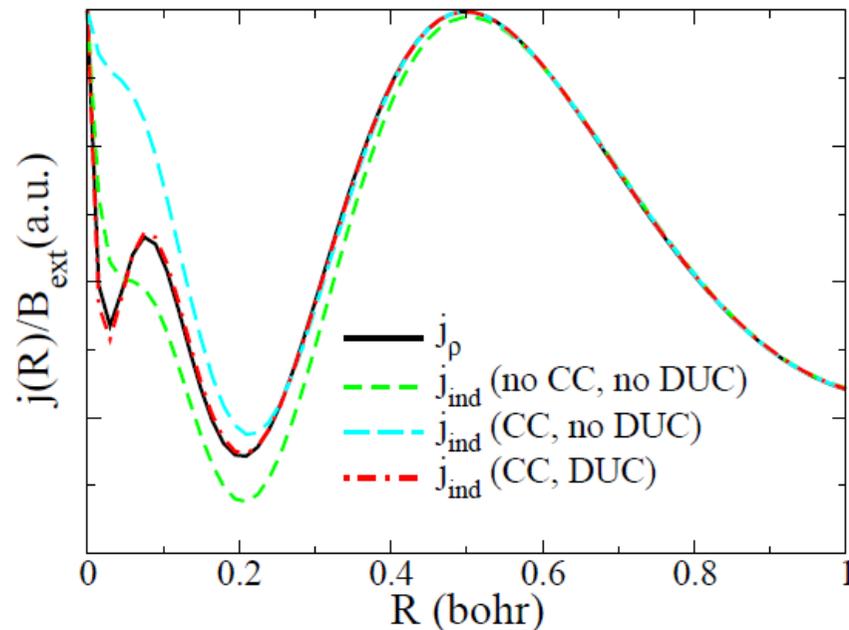
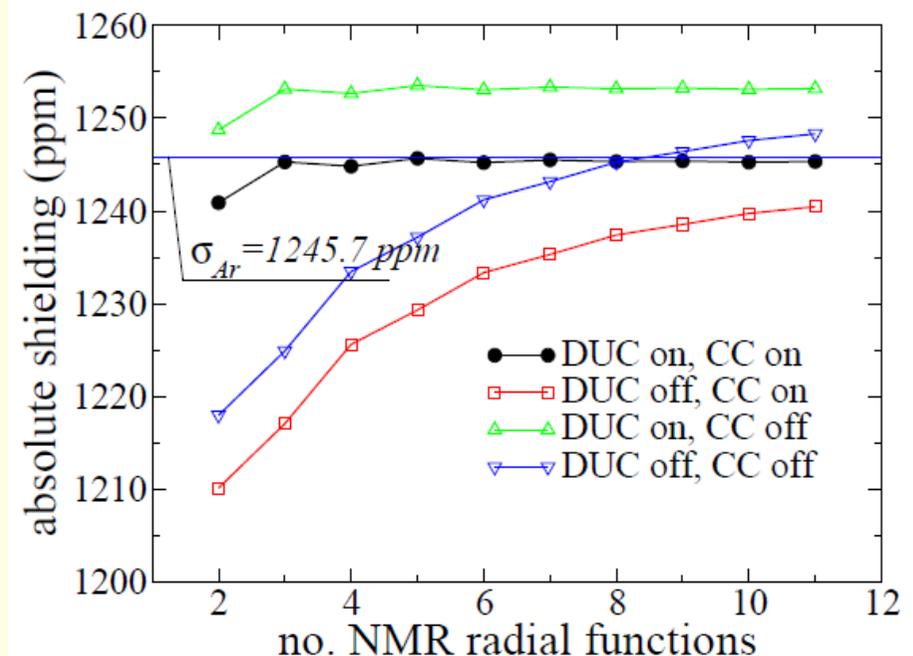


FIG. 1: Comparison of the induced current computed for an Ar atom with and without DUC and CC corrections (see text). For clarity only the valence 3s and 3p contribution is shown.  $j_\rho$  is the diamagnetic current calculated with the spherical charge density of 3s and 3p states using Eq.18.

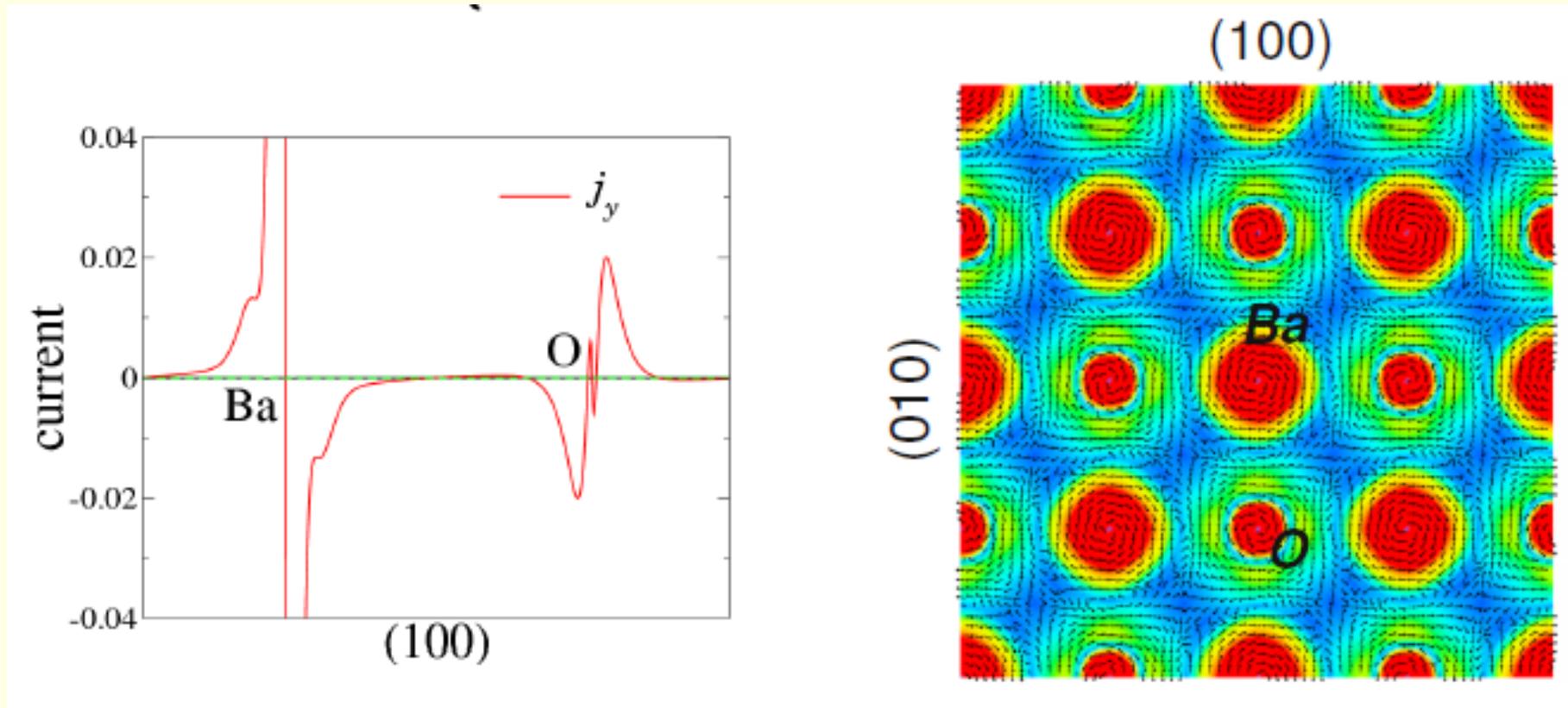




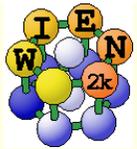
# Induced current in LAPW



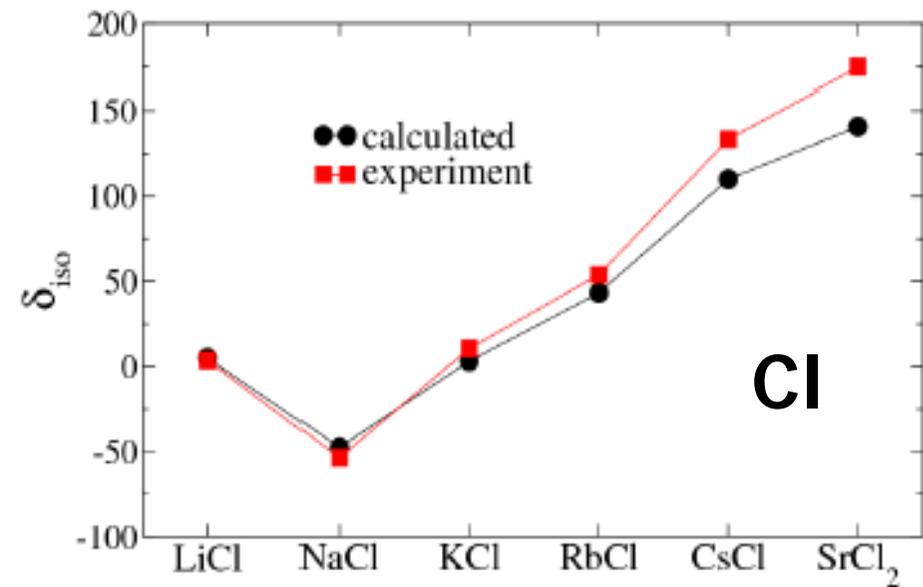
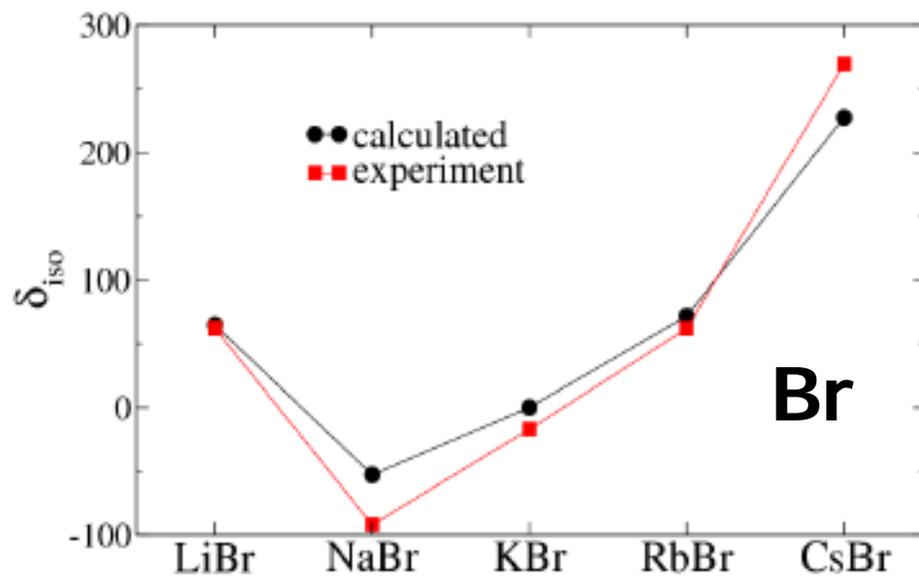
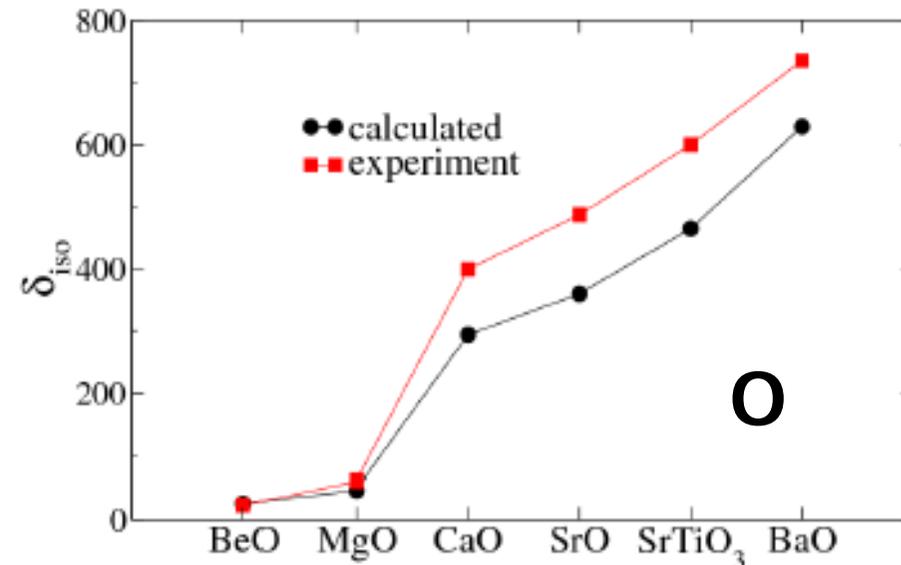
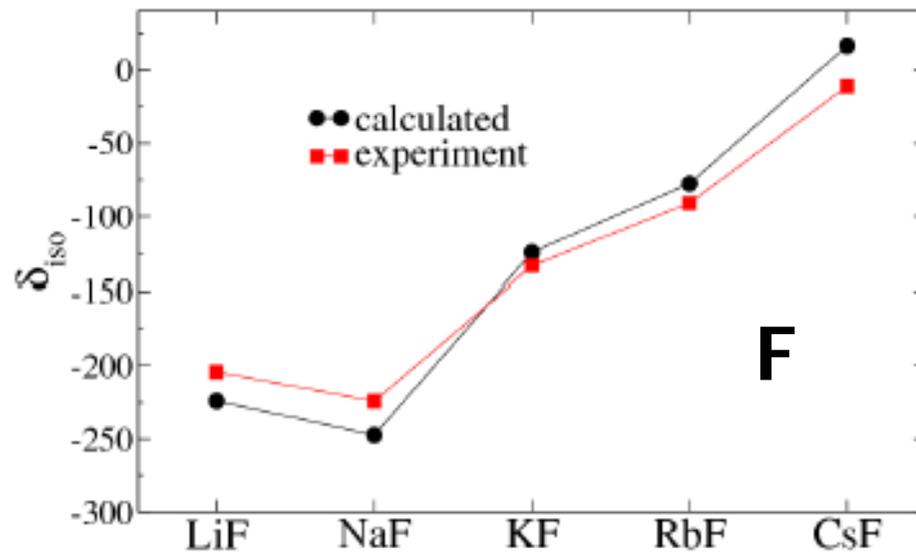
Induced current field for BaO (fcc) ,  $B_{ext}$  in (001)



$$\mathbf{j}_{ind}(\mathbf{r}) = \begin{cases} \sum_{\mathbf{G}} \mathbf{j}_{\mathbf{G}} e^{i\mathbf{G}\cdot\mathbf{r}}, & \mathbf{r} \in I \\ \sum_{l,m} \mathbf{j}_{l,m}^{\alpha}(r) Y_{l,m}(\hat{r}). & \mathbf{r} \in S_{\alpha} \end{cases}$$



# NMR shifts for F, O, Br, Cl





# Interpretation of $^{19}\text{F}$ NMR shielding in alkali fluorides

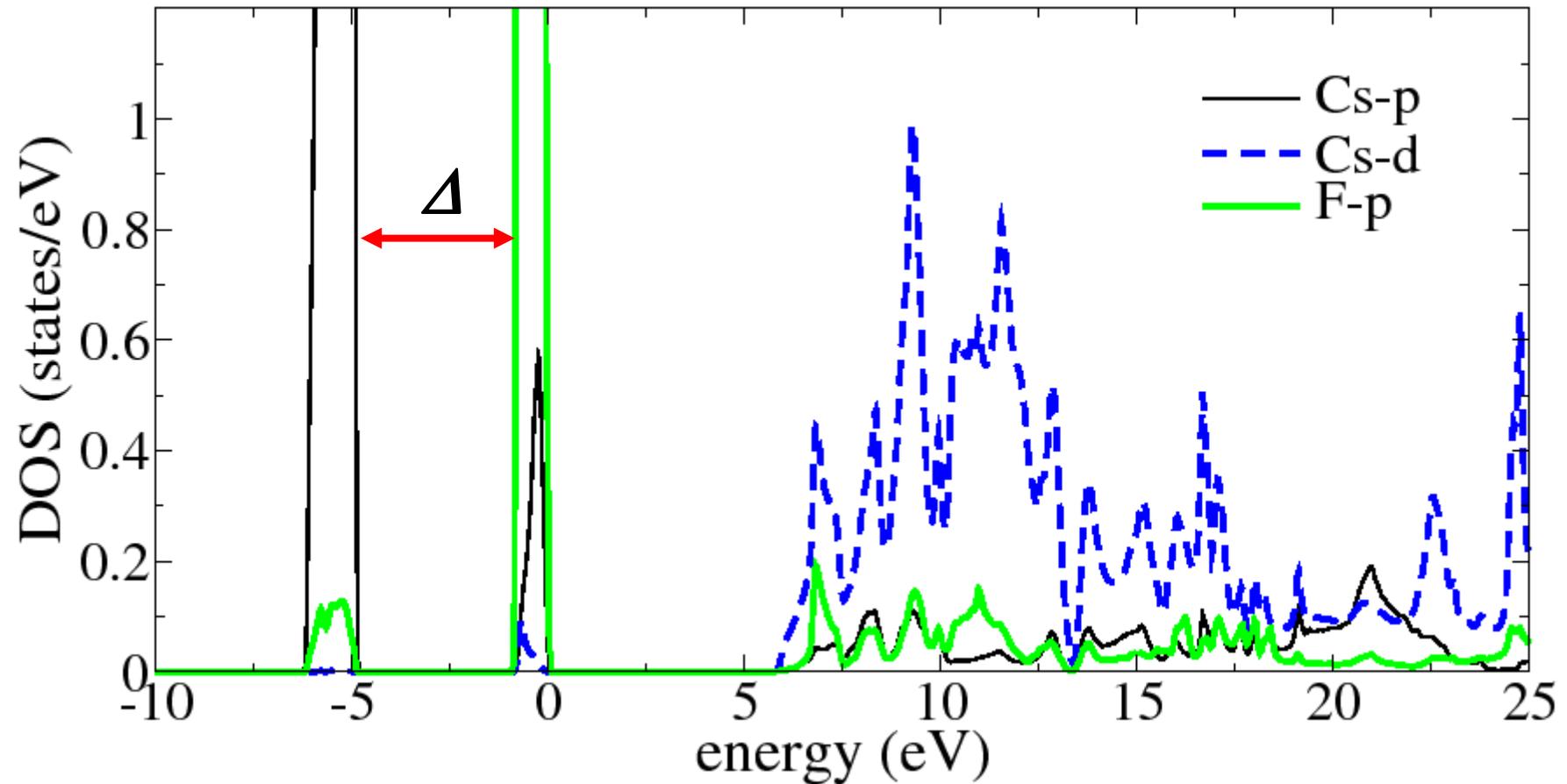


- band wise analysis
- character analysis (s,p,d) of the wave function of occupied and unoccupied states



# DOS of alkali fluorides (CsF)

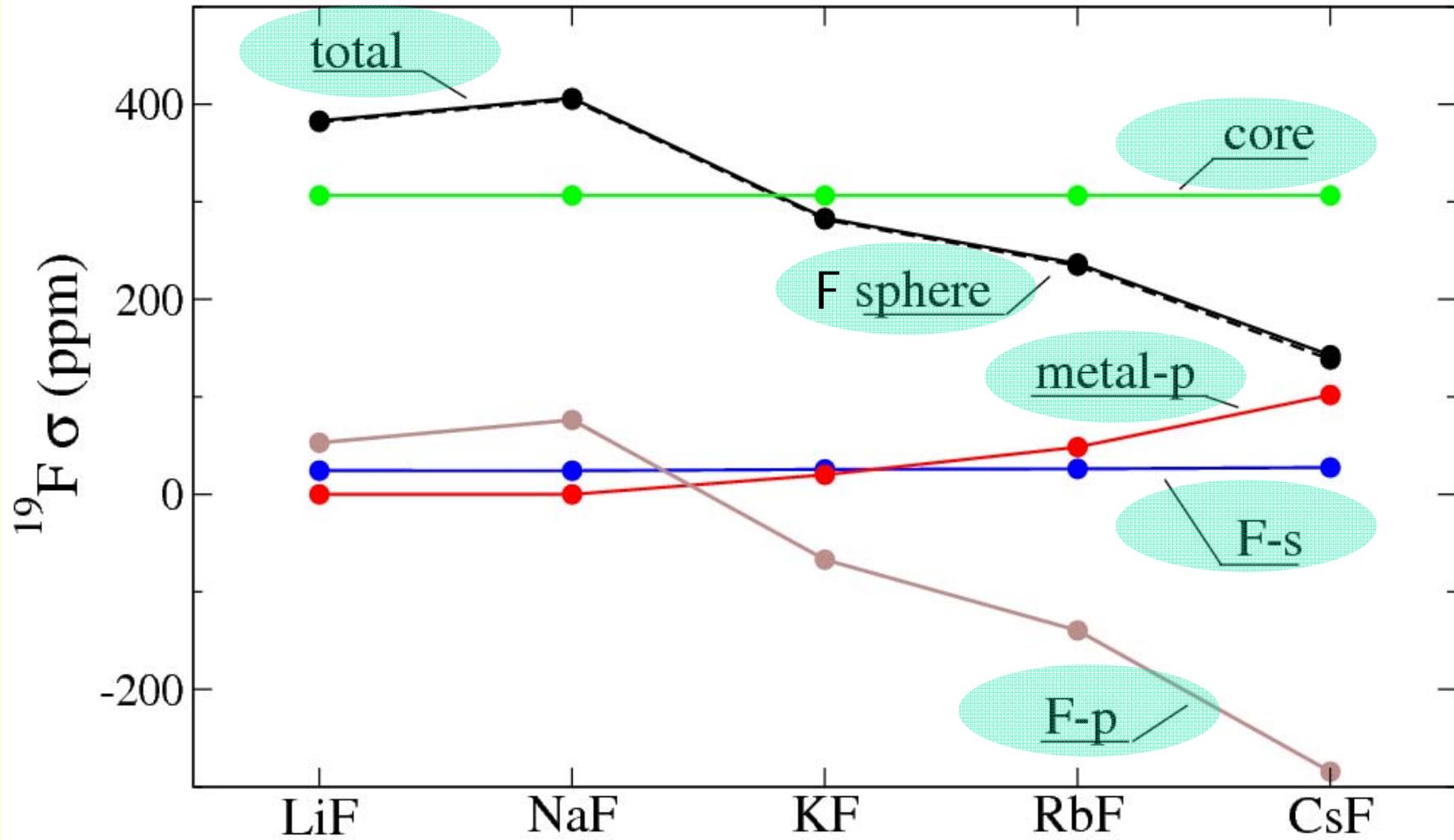
metal-p      F-p band



$\Delta$  varies between 5 eV for CsF to 20 eV for NaF



# Band wise analysis of the isotropic shielding in MF





# Decomposition of NMR shift



- decomposition of NMR shift according to  $s, p, d$  - character and atom

- $$\Psi_0 = \sum_{at} \sum_{lm} R_{at,lm} Y_{lm}$$

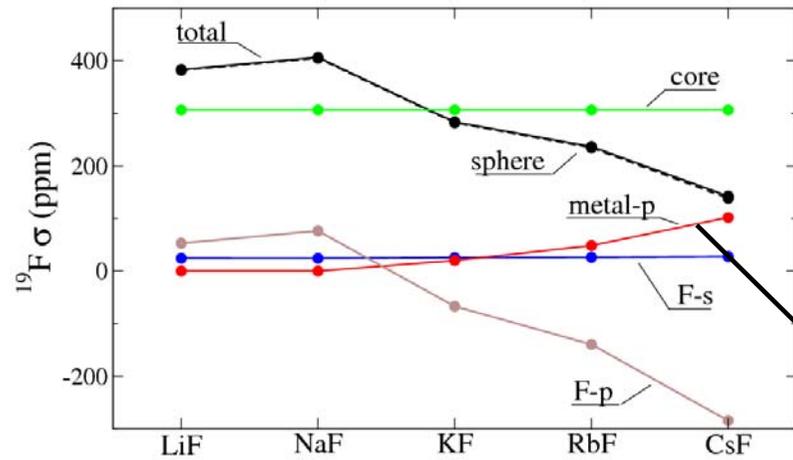
- decomposition according to ground state  $\Psi_0^{(0)}$  and perturbed states  $\Psi_0^{(1)}$

$$\mathbf{j}_{ind}(\mathbf{r}') = \frac{1}{c} \sum_o \Re \left[ \langle \Psi_0^{(0)} | \mathbf{J}^p(\mathbf{r}') G(\epsilon_o)(\mathbf{r}-\mathbf{r}') \times \mathbf{p} \cdot \mathbf{B} | \Psi_0^{(0)} \rangle \right]$$

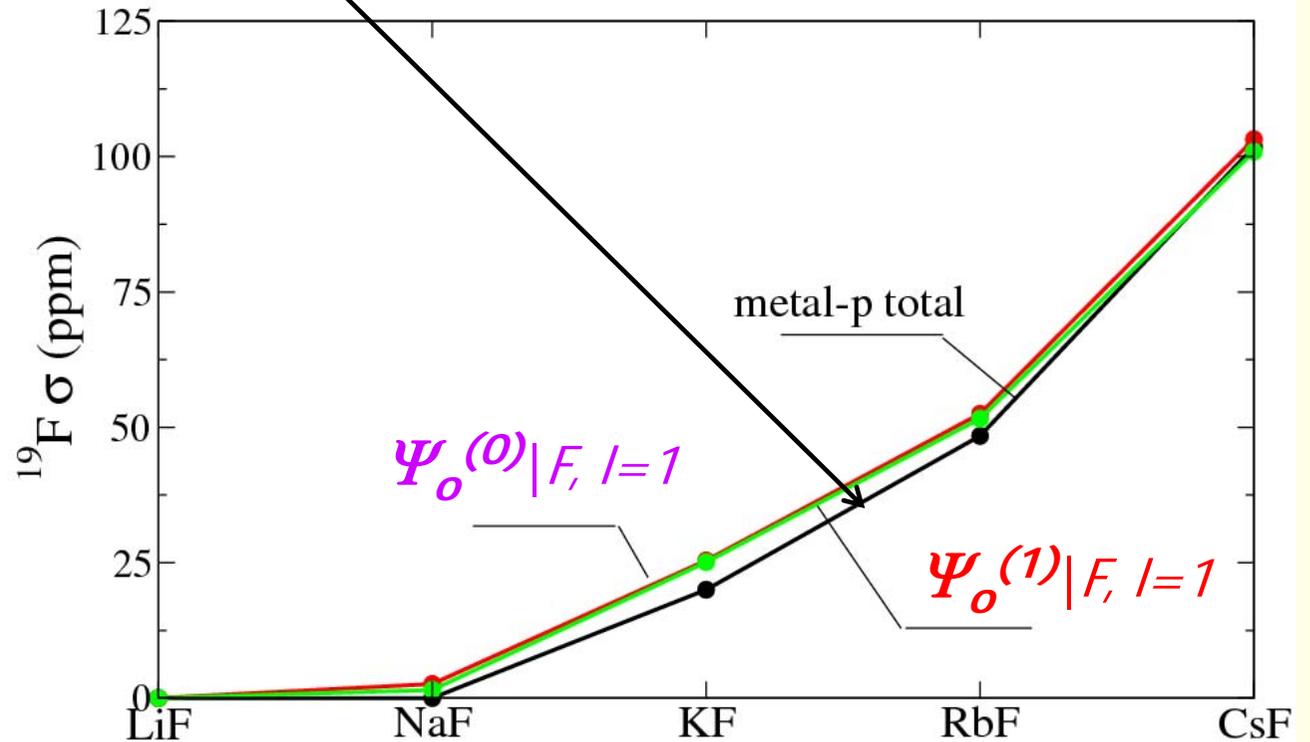
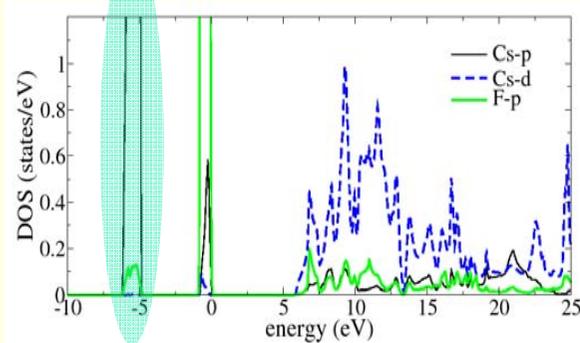
$\Psi_0^{(1)}$



# metal-p band contribution

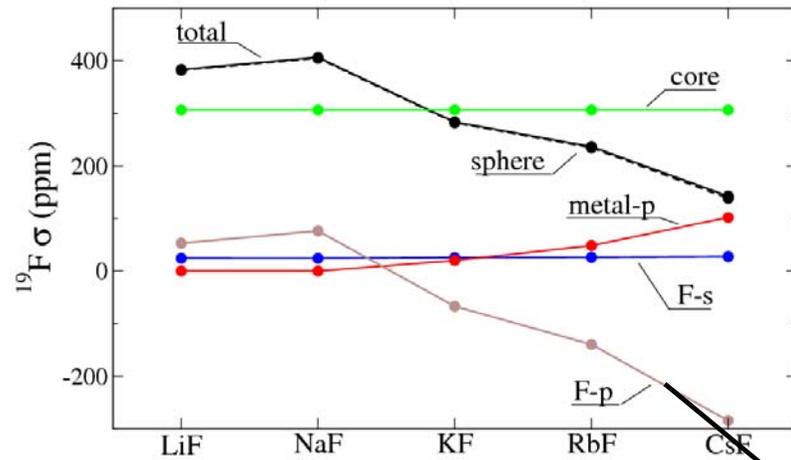


$$j_{ind}(\mathbf{r}') = \frac{1}{c} \sum_o \Re [\langle \Psi_o^{(0)} | \mathbf{J}^P(\mathbf{r}') | \Psi_o^{(1)} \rangle]$$

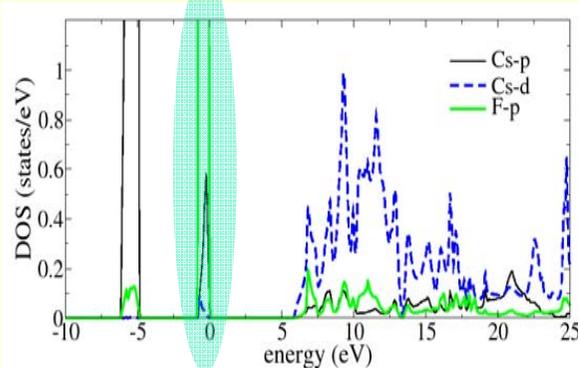
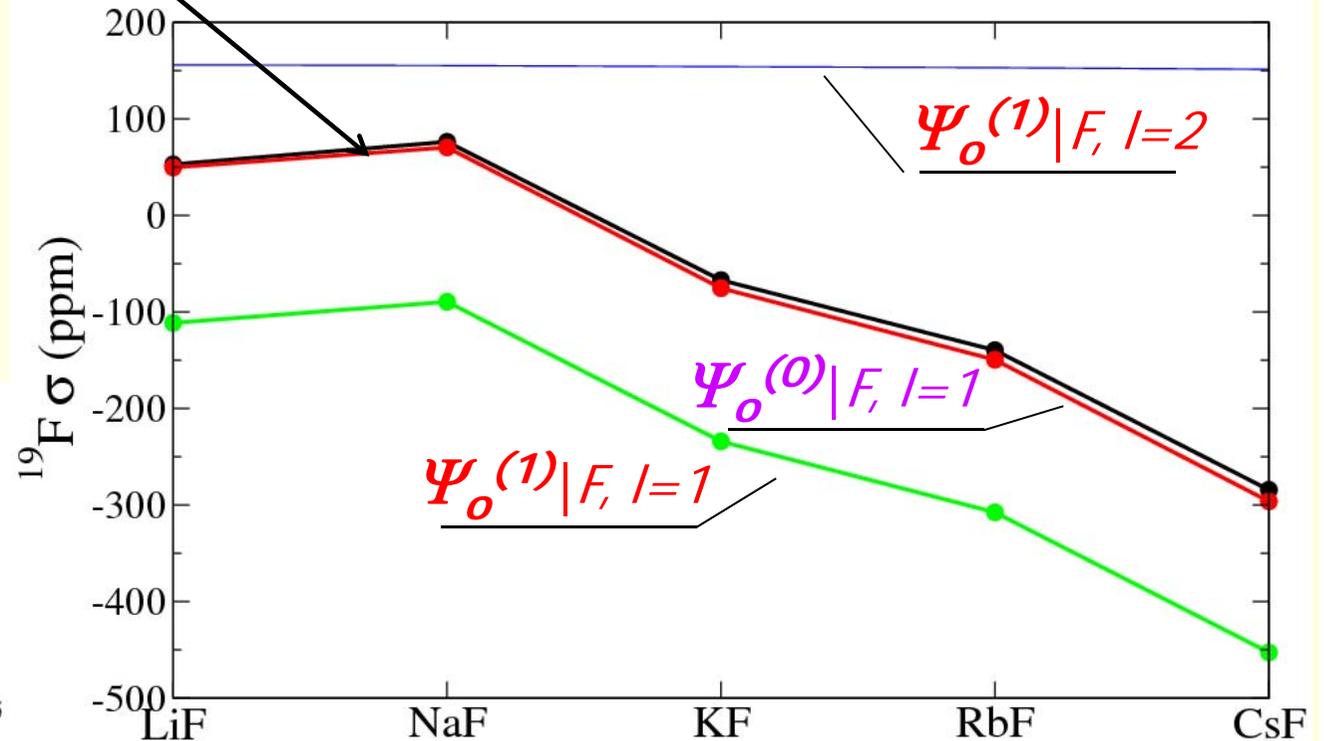


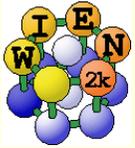


# F-p band contribution

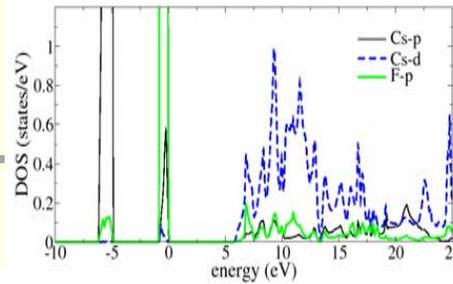


$$j_{ind}(\mathbf{r}') = \frac{1}{c} \sum_o \Re [\langle \Psi_o^{(0)} | \mathbf{J}^P(\mathbf{r}') | \Psi_o^{(1)} \rangle]$$

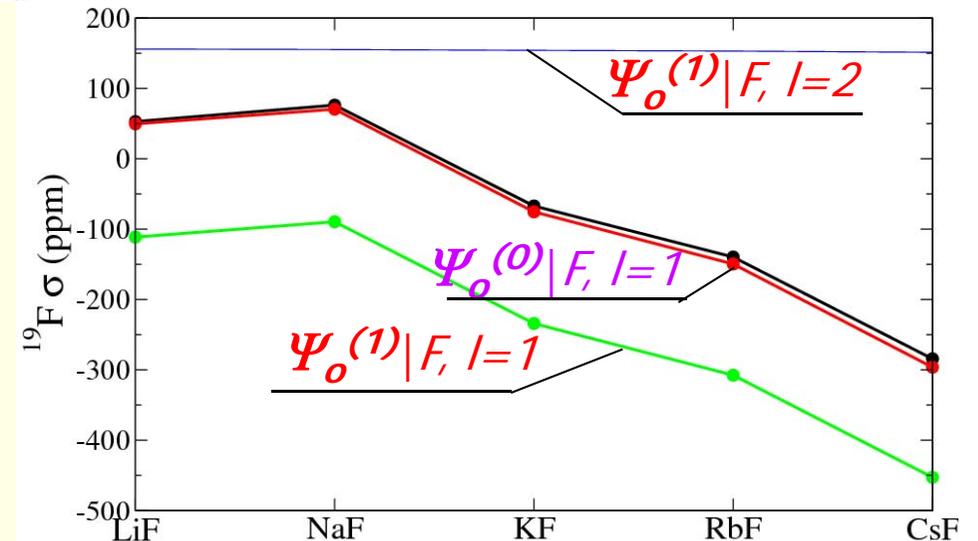
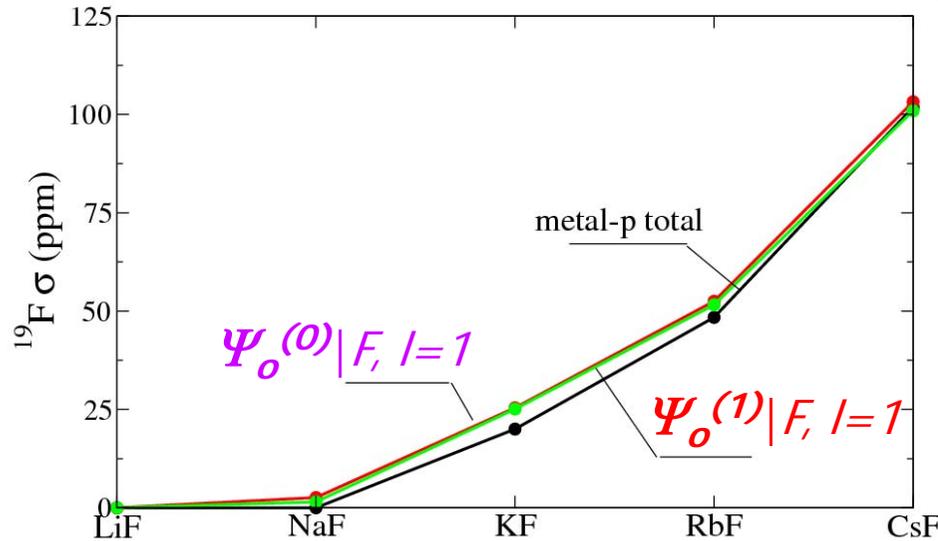




## metal p band



## F-p band



the only important ground state contribution

$$\Psi_0^{(0)} | F, l=1$$

the only important ground state contribution

**positive, increasing**  
contribution within the series

$$\Psi_0^{(1)} | F, l=1$$

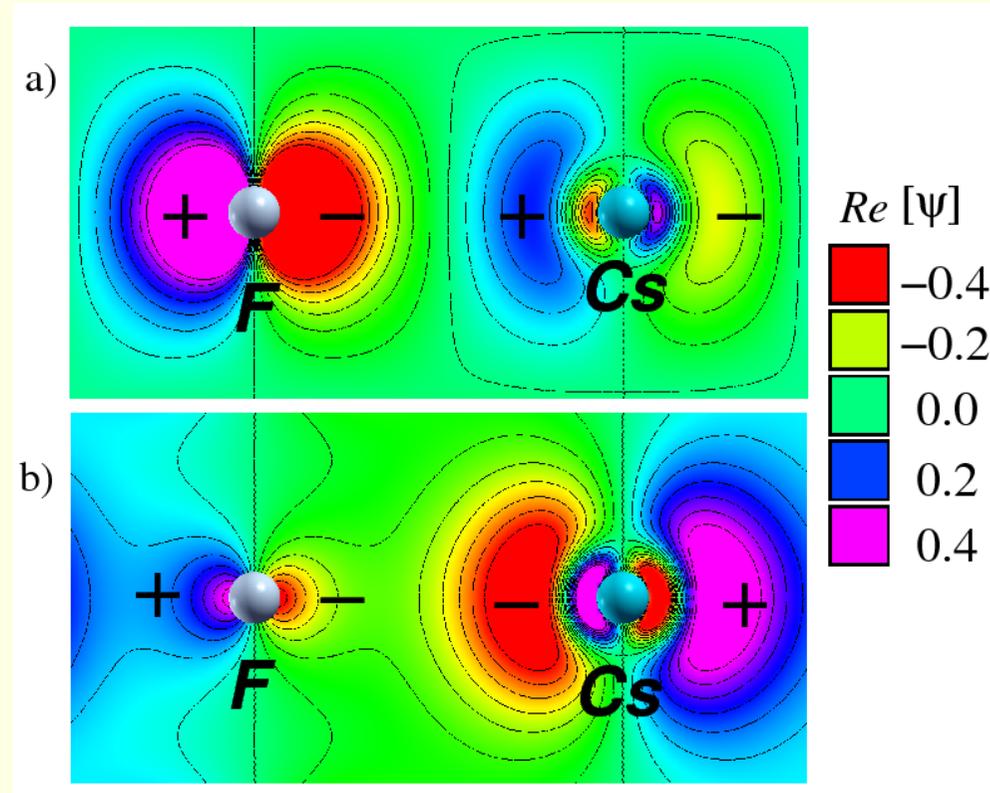
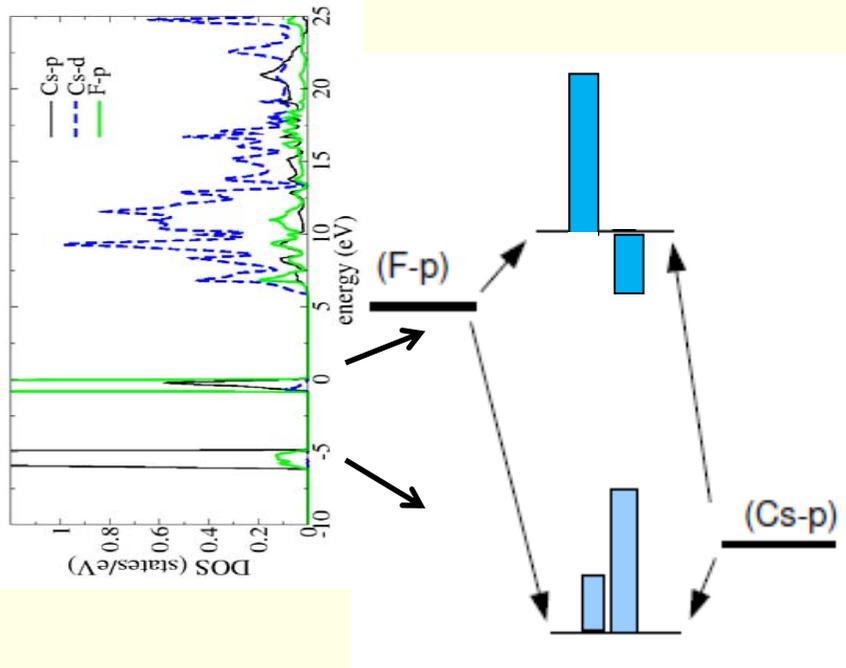
**negative, decreasing**  
contribution within the series

$$\Psi_0^{(1)} | F, l=2$$

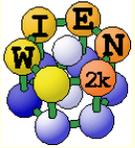
constant contribution  
within the series



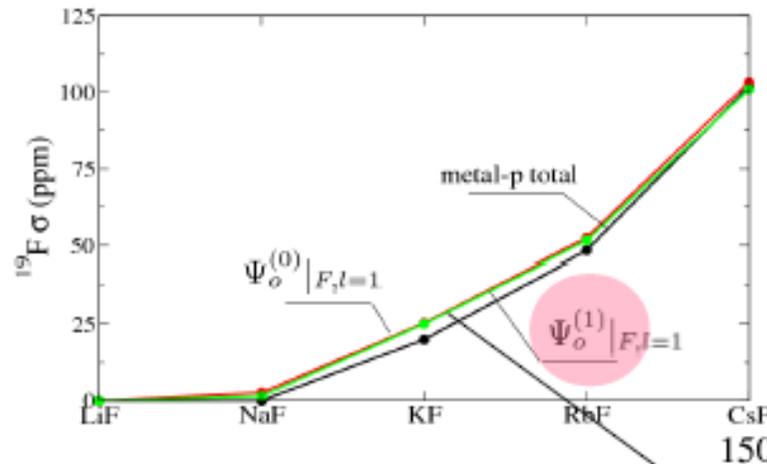
## Re[ $\Psi$ ] at X-point of CsF



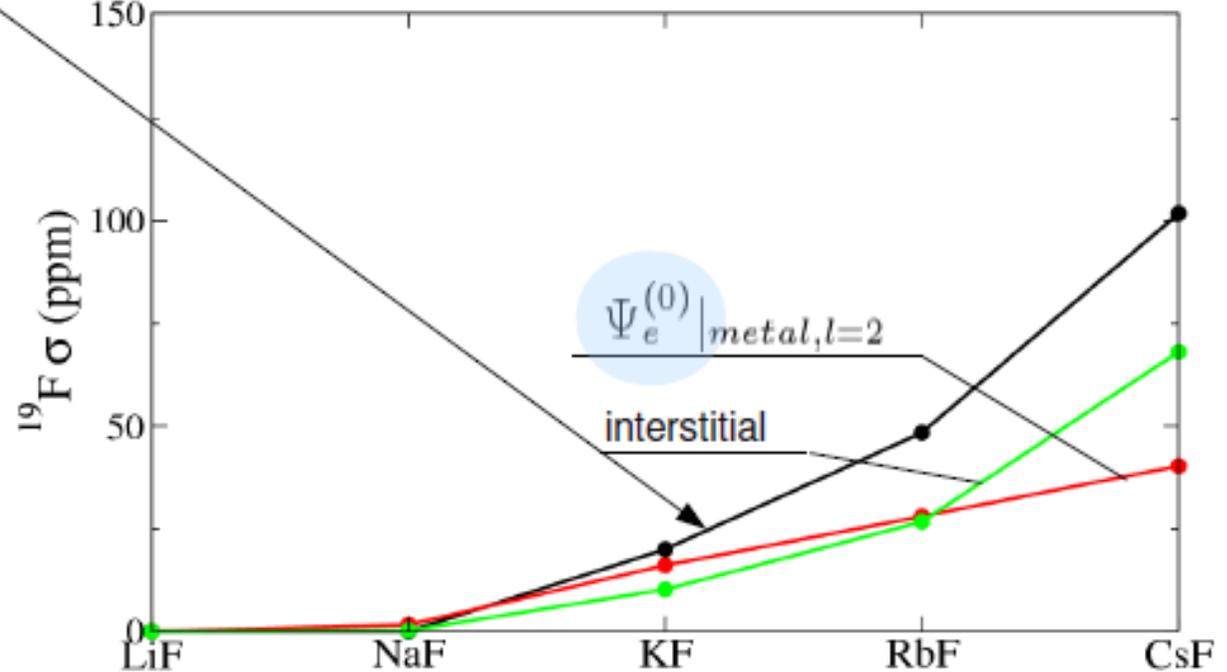
- a) F-p band, **anti-bonding** character of the Cs-p and F-p orbitals, **negative** contribution to the shielding
- b) Cs-p band, **bonding** character between Cs-p and F-p orbitals, **positive** contribution to the shielding.

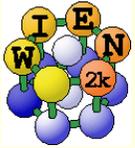


# metal-p band contribution

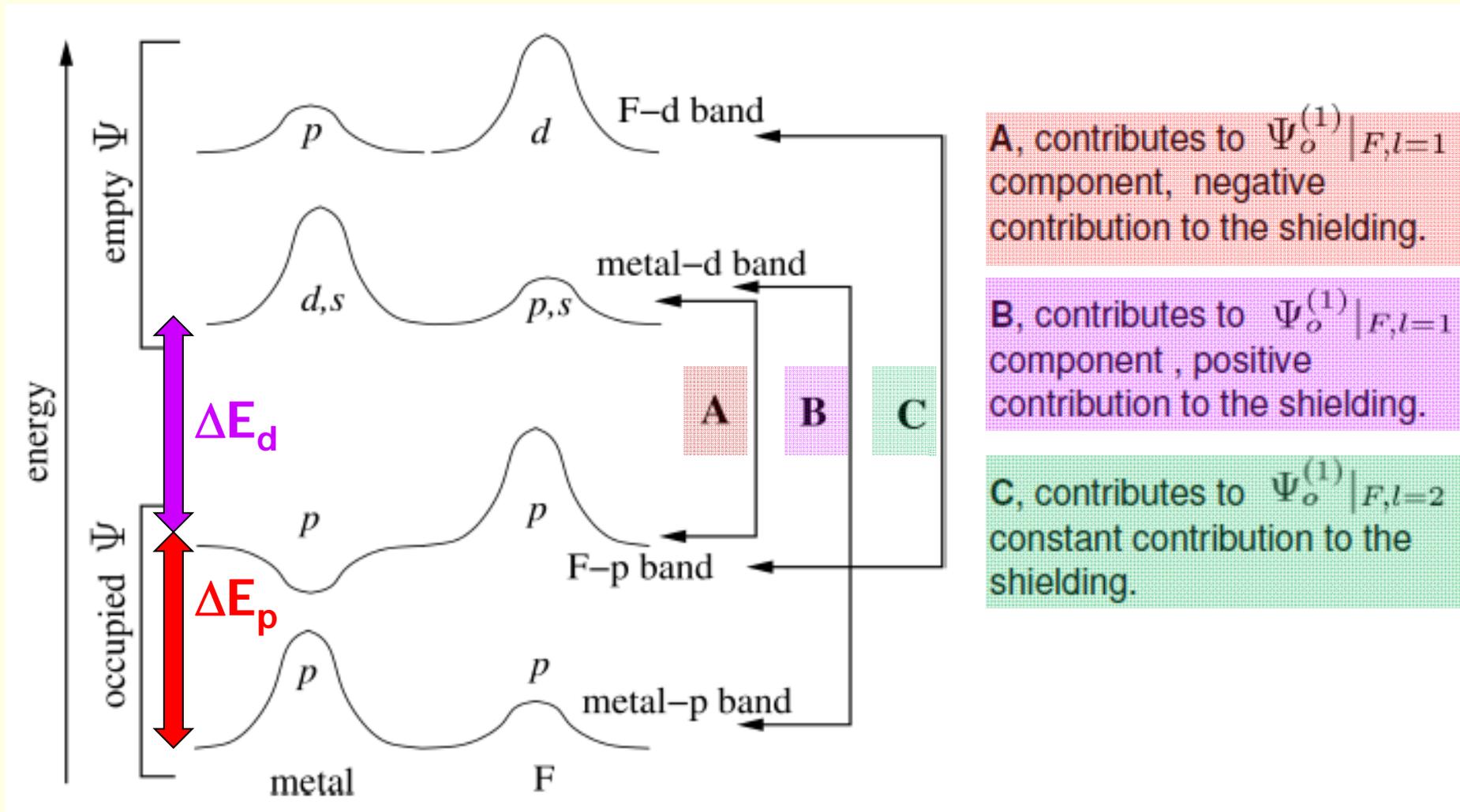


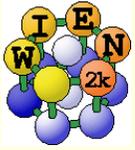
$$|\Psi_o^{(1)}\rangle = \sum_e |\Psi_e^{(0)}\rangle \frac{\langle \Psi_e^{(0)} | H^{(1)} | \Psi_o^{(0)} \rangle}{\epsilon - \epsilon_e}$$





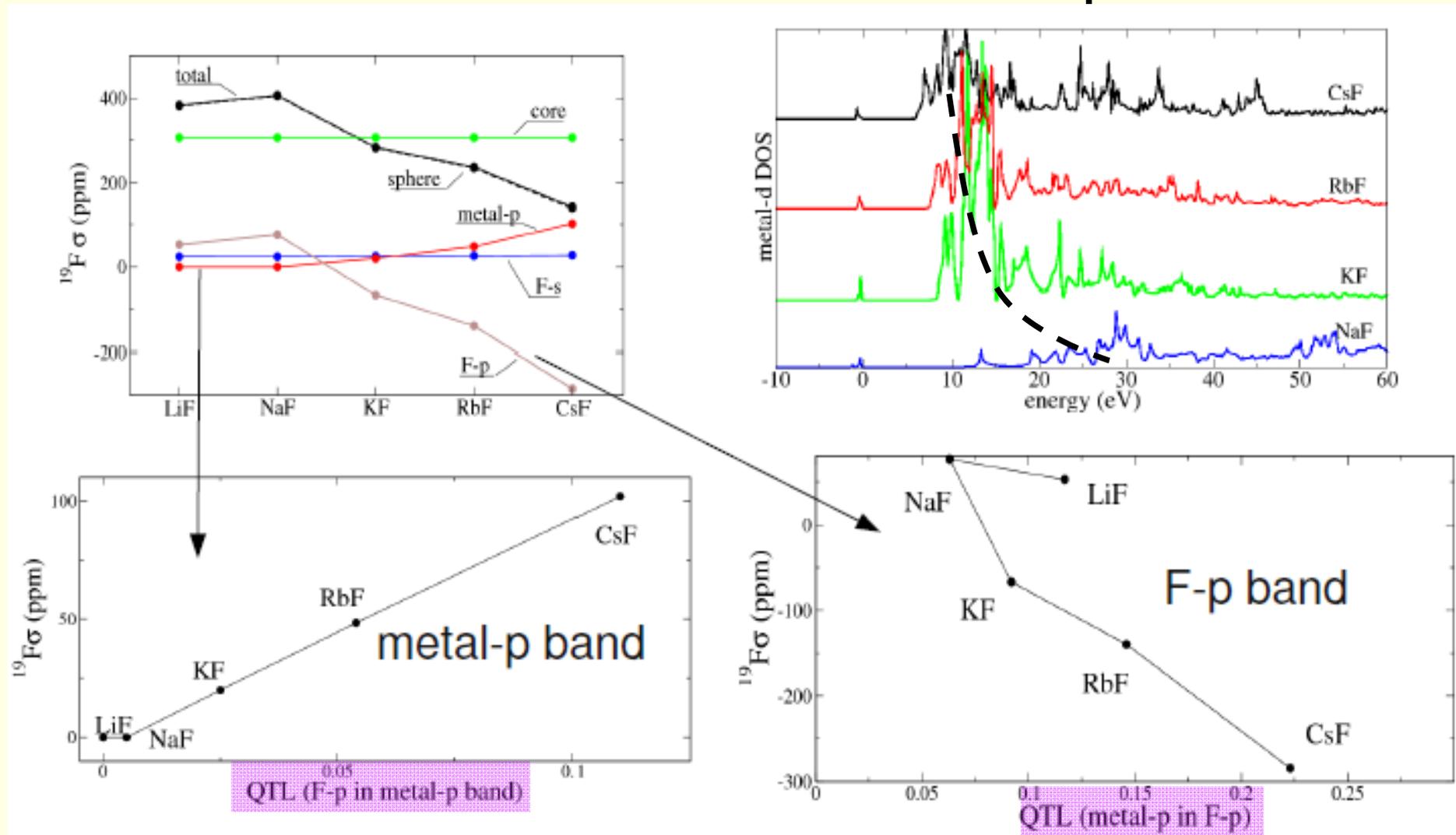
# Interactions relevant for NMR chemical shifts in alkali fluorides





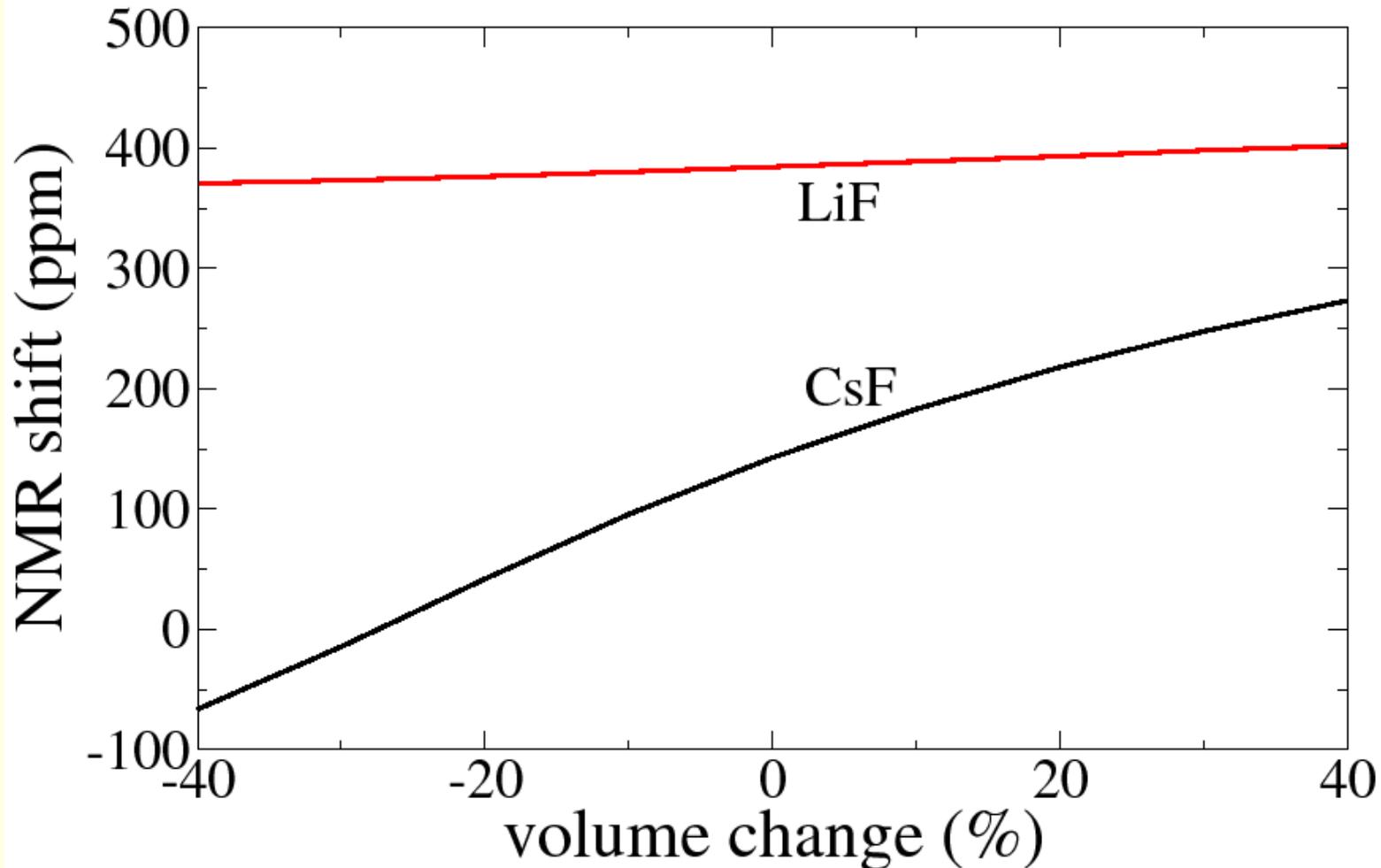
# A, B interactions:

coupling to the metal-d states, due to F-p – metal-p hybridization  
d-band position





## Effect of bond distance on the shielding



- decreasing volume leads to stronger Me-p F-p interaction and to more negative shielding (Li does not have "Li-p band")

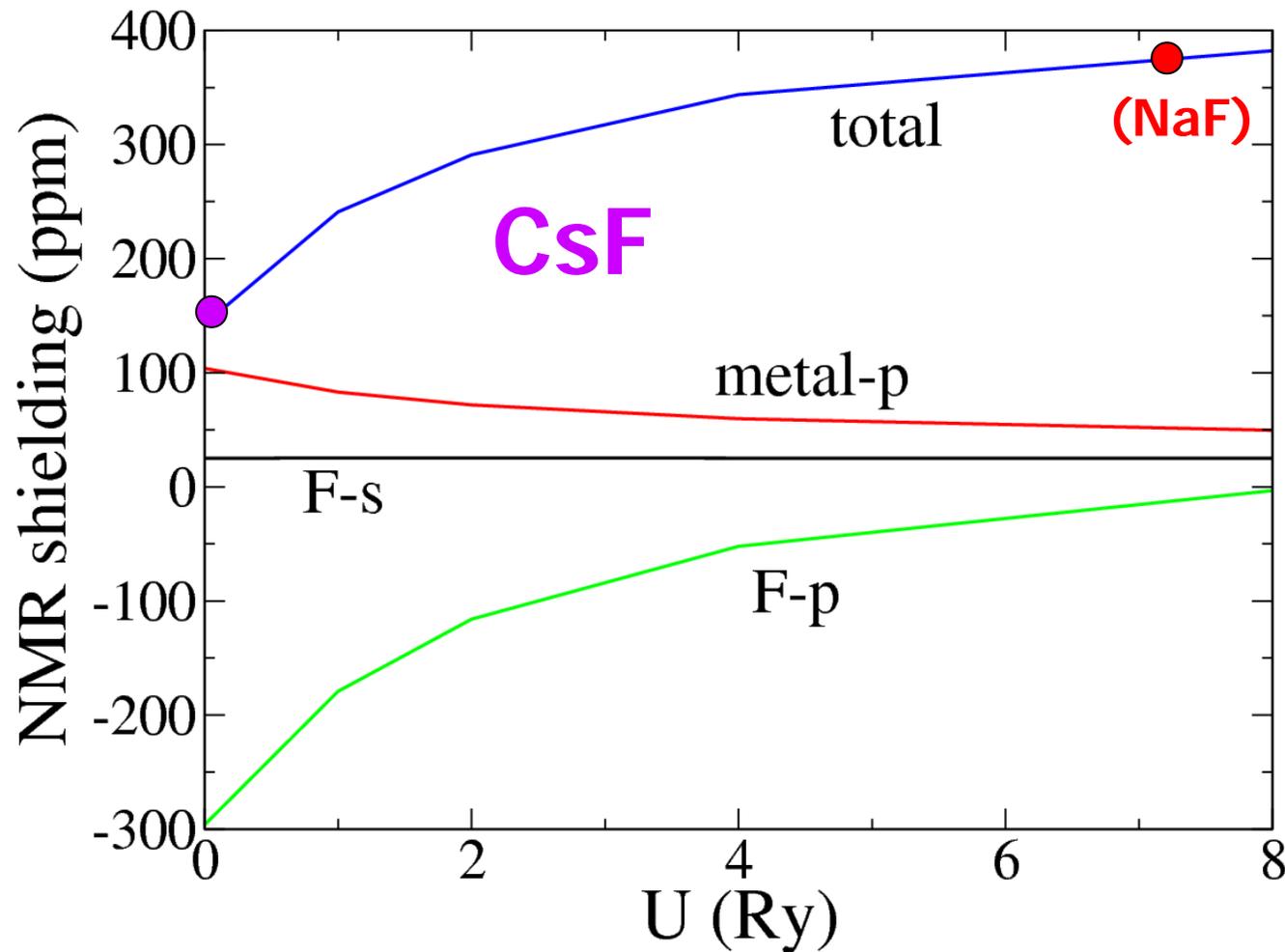


# Effect of position of metal-d band on the F shielding



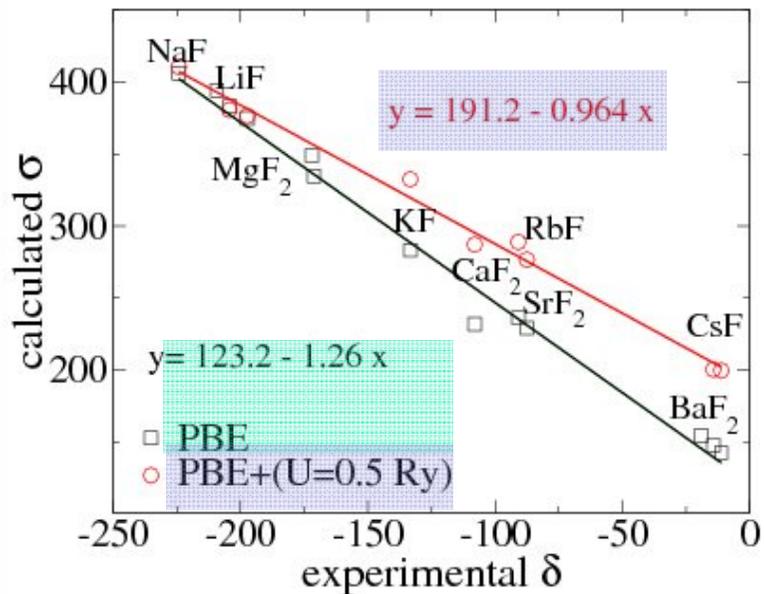
## LDA+U acting on Cs-d

$$|\Psi_o^{(1)}\rangle = \sum_e |\Psi_e^{(0)}\rangle \frac{\langle \Psi_e^{(0)} | H^{(1)} | \Psi_o^{(0)} \rangle}{\epsilon - \epsilon_e}$$





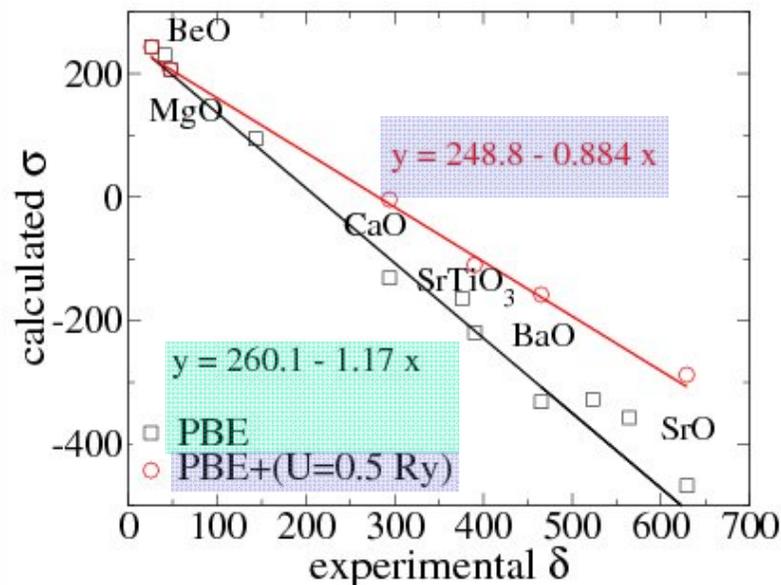
# The "slope" - problem



exp.  $\delta$  vs. theoretical  $\sigma$ :  
The slope must be **ONE**

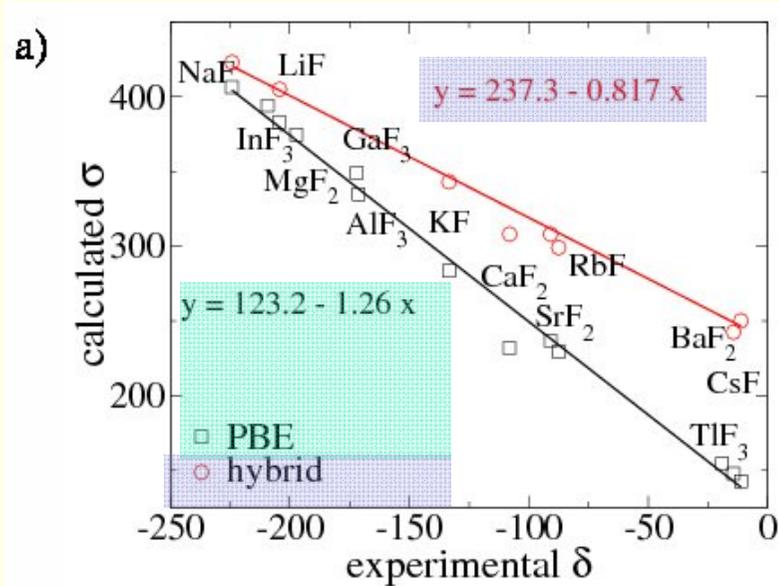
**PBE: slope is too big**

**PBE+U (metal d-states):**  
with **one** U value it is **not**  
possible to fix **oxygen AND**  
**fluorine CS.**



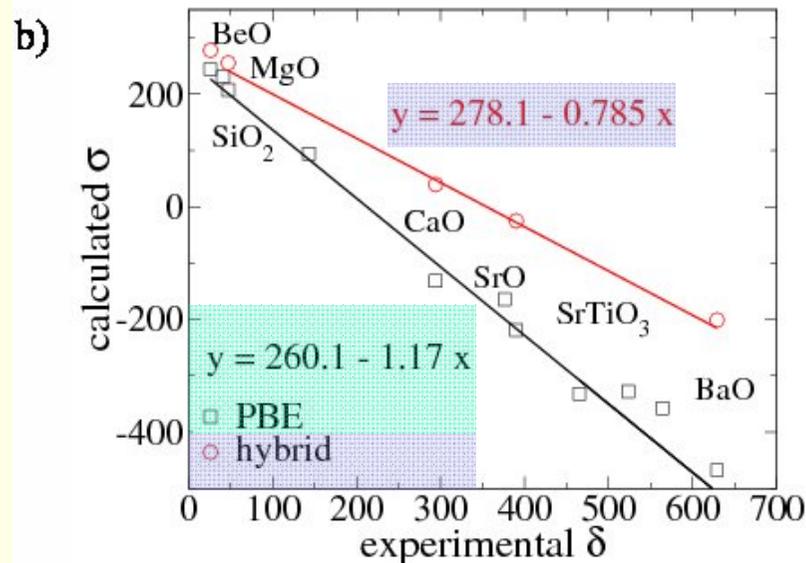


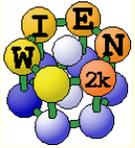
# the slope - problem



- hybrid-DFT is the standard method in CS calculations of molecules (Gaussian)

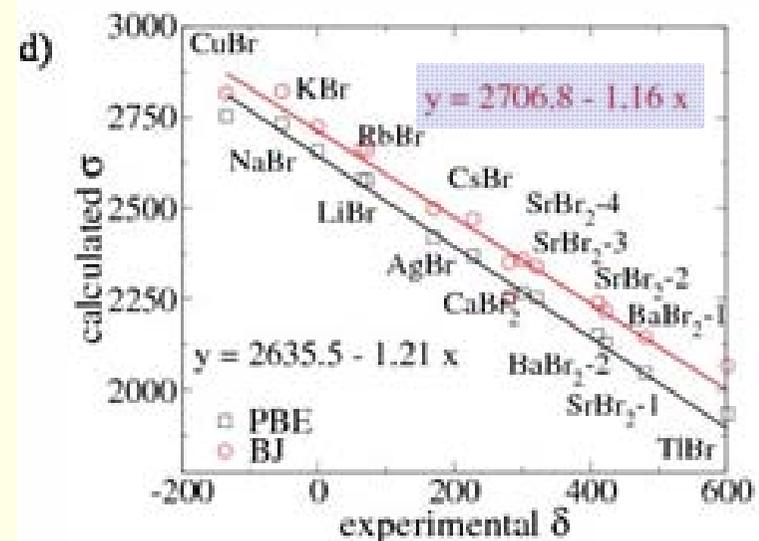
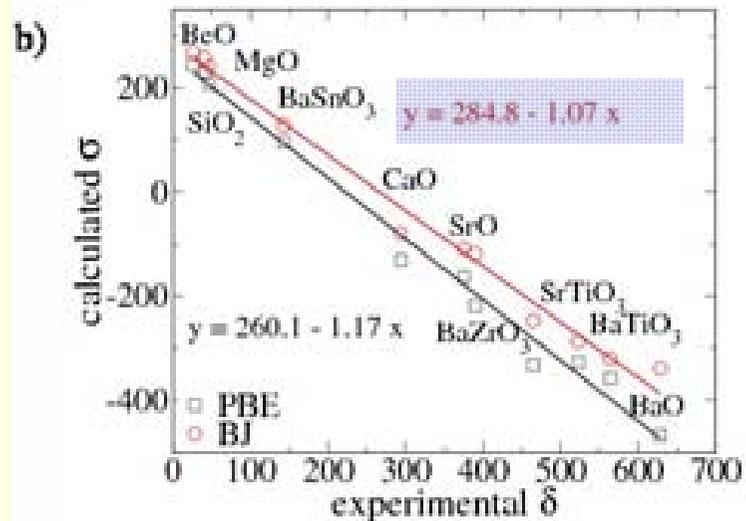
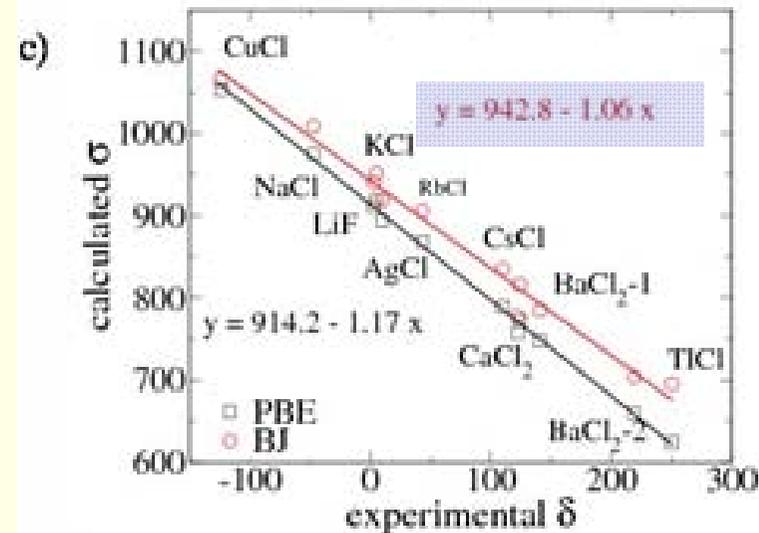
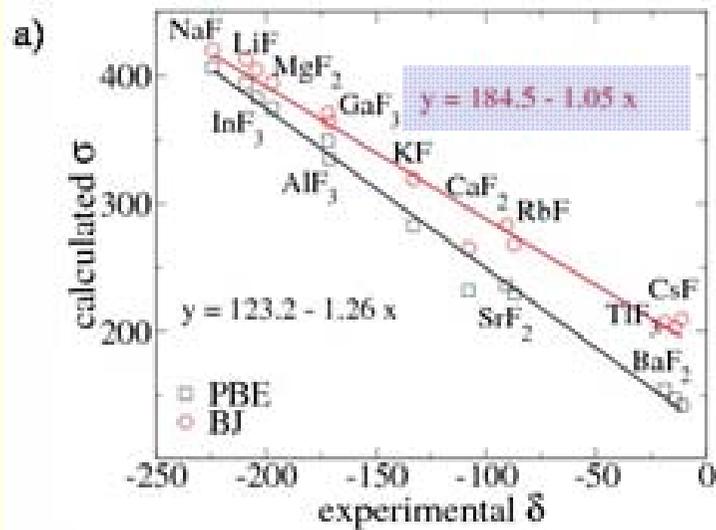
- for (ionic) solids YS-PBE0 (HSE) gives a much **too large** correction (smaller mixing ??)





# the slope - problem

BJ-potential (OEP) seems quite reasonable for ionic compounds



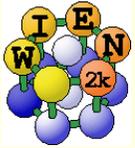


# Summary:



## NMR chemical shifts:

- shielding of anions in solids determined by:
  - *strength of metal-p -- F-p hybridization*
    - distance of metal-p band from anion-p band
    - bond distance, number of neighbors
  - *position of empty metal-d states*



# Acknowledgement



**Robert Laskowski (TU Vienna)**

NMR: PRB **85**, 035132 (2012)  
PRB **85**, 245117 (2012)



**Thank you for your attention !**